

Purification and physicochemical characterization of exopolysaccharide produced by a novel brown-rot fungus *Fomitopsis meliae* AGDP-2

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ABSTRACT

The present study represents the purification and characterization of exopolysaccharide produced by *Fomitopsis* meliae AGDP-2. The water-soluble EPS obtained was purified using Sephadex G-75 column and characterized for its physical and biochemical properties. HPTLC and HPLC analysis indicated that EPS is composed of glucose and mannose residues. Many functional groups such as carboxyl, amide, hydroxyl, and glycosidic linkages were detected by FT-IR spectroscopy. Furthermore, glucose and mannose were detected from NMR spectra with α -(1 \rightarrow 6) linkages, thereby signifying the heteropolymeric nature. Using GPC, molecular weight of EPS was estimated to be 2.48 × 10⁶ Daltons. The SEM-EDX and AFM analysis revealed that EPS is compact and possesses cube-like structural units and nano-cracks with elemental composition of O, K, Na, Mg, S, Cl, and Ca. While studying its rheological behavior, EPS was found to be pseudoplastic in nature. As the physical properties and biochemical composition influence the bioactivities; the purified EPS from *F. meliae* AGDP-2 can be further evaluated for various bioactive properties such as antioxidant, anti-cancer, anti-inflammatory, antidiabetic, antibacterial, antibiofilm, as well as antiviral properties.

1. INTRODUCTION

Exopolysaccharides (EPSs) are polymers made up of monosaccharides that are linked through glycosidic bonds. They have high molecular mass and are secreted by plants, animals, and microorganisms. Fungi and bacteria have been prioritized over other microorganisms for their excellent exopolysaccharide production capabilities [1]. Chemically, EPSs are primarily composed of the carbohydrate moieties. In addition to carbohydrate molecules, inorganic and organic functional groups such as acetate, amine, phosphate, succinate, and sulfate have also been found in their chemical structures [2]. On the basis of their characteristics, these compounds can be employed as an alternative product of plants and algae being considered as non-hazardous and naturally biodegradable products [3].

Apart from bacteria, yeasts, lower filamentous fungi, and higher basidiomycetes are also known to produce EPS. Even though bacteria are regarded as the most common producers of EPS, the extraction procedure of EPS produced by yeasts and fungi is convenient and quick as compared to bacterial EPSs; and thus, it can be directed to a larger scale [4]. Fungi have been reported to be a promising source of pharmaceutically important compounds. The polysaccharides produced by fungi have been regularly investigated for their biological

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activities. The production of polysaccharides occurs in different phases of the fungal life cycle and can be found intracellular, attached to the cell, or secreted in the extracellular environment.

One of the members of basidiomycetes fungi, *Fomitopsis pinicola*, is commonly used as a medicinal mushroom in Asia [5]. Although there are a number of reports on properties EPS from *Fomitopsis* fruiting body, few studies have been carried out on the structural and morphological characteristics of EPS from *Fomitopsis*. Hsieh *et al.* [6] reported that there can be noteworthy differences in the structure or composition of polysaccharides from various sources. To the best of the author's knowledge, no reports have been made to date on the compositional characteristics of EPS from *Fomitopsis meliae*.

Given the limited knowledge of the EPS synthesized by *F. meliae*, the present study was conducted to purify and characterize the EPS produced under submerged cultivation of *Fomitopsis meliae* AGDP-2. The characterization of EPS including monosaccharide composition (high-performance thin-layer chromatography [HPTLC] and HPLC), molecular weight (GPC), rheological behavior, surface morphology (SEM and AFM), as well as IR and nuclear magnetic resonance (NMR) spectra was thoroughly studied.

2. MATERIALS AND METHODS

2.1. Chemicals and Reagents

The monosaccharide standards (glucose, mannose, arabinose, xylose, lactose, galactose, mannitol, sucrose, sorbitol, and xylitol) used for

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reference in HPTLC and HPLC as well as dextran standard used as reference in GPC were purchased from Sigma-Aldrich, St. Louis, USA. All other analytical grade chemicals were purchased from HiMedia, Mumbai, India.

2.2. Microorganisms and Growth Conditions

The native fungal isolate *F. meliae* AGDP-2 was maintained on PDA medium. For the production of EPS, six 8 mm diameter sized mycelial agar plugs were aseptically inoculated in 100 mL of statistically optimized Tien and Kirk medium (g/L: Glucose 37.5; peptone 6.0; yeast extract 2.0; CaCl₂ 0.4; KH₂PO₄ 2.0; MgSO₄ 0.9; and pH 5.3) and were incubated at 28°C for 10 days.

2.3. Extraction of EPS and Biomass Determination

The dry biomass content was determined separating the biomass from fermentation broth using muslin cloth followed by drying at 60°C. EPS precipitation and extraction were achieved by adding ice-cold isopropyl alcohol to the filtrate in 1:1 v/v proportion. The crude extracted EPS was dried at 50°C to estimate the yield.

2.4. Purification of EPS

The precipitated EPS was dialyzed against distilled water overnight at 4°C. The EPS obtained was further subjected to Sephadex G-75 Gel Permeation Chromatography column. The fractions were collected at the flow rates of 0.2 mL/min. Total carbohydrate and protein contents of crude, dialyzed and purified EPS were measured.

2.5. Characterization of EPS

2.5.1. Total carbohydrate and protein content analysis

The total carbohydrate content of EPS was measured by phenolsulfuric acid method using D-glucose as a standard [7]. The total protein content of crude and partially purified EPS was measured by Folin-Lowry's method. Bovine serum albumin was used as a standard [8].

2.5.2. Monosaccharide composition analysis

Primarily, the presence of sugar monomers in EPS was determined using HPTLC. In brief, EPS was hydrolyzed in 4 M trifluoroacetic acid (TFA) in a sealed tube at 100°C for 4 h. After hydrolysis, sample was cooled to room temperature and TFA was coevaporated with methanol to dryness under vacuum. The hydrolyzed EPS sample and standard sugars (glucose, maltose, xylose, and mannose) were run on silica gel 60 F254 plate (Merck, Germany). A sample of 10 µl was spotted on the TLC plate using micro syringe (HPTLC, CAMAG, Linomat 5). The mobile phase was composed of the ternary solvent system butanol/ ethanol/H₂O in the ratio of 5:3:2 v/v/v. After drying, the TLC plate was developed using methanol/H₂SO₄ developing solution (9:1 v/v). On heating the TLC plate above 100°C in hot air oven for about 5 min, the sugar spots were visually observed. Chromatogram was observed using Camag TLC scanner 3.

The hydrolyzed EPS was applied to HPLC with RI detector system (Shimadzu LC-2010C HT) having Aminex® HPX 87 H Ion Exclusion column (300 mm \times 7.8 mm). The column temperature was maintained to 65°C temperature. A 5 mM sulfuric acid was used as a mobile phase to achieve separation of sugar monomers.

2.5.3. UV-visible spectroscopy

The homogeneity of EPS was checked using Thermo Scientific Multiskan Go spectrophotometer system. The EPS sample was

scanned at the wavelength ranges of 200–1000 nm with the interval of 1 nm.

2.5.4. Fourier-transform infrared (FT-IR) analysis

To study the bonding patterns and identify the major functional groups, the EPS was subjected to FT-IR analysis using a PerkinElmer (Spectra GX) spectrometer. The solid EPS sample was pressed into KBr pellets. The IR spectrum was recorded between the frequency ranges of 4000- 400 cm^{-1} at a resolution of 200 cm⁻¹.

2.5.5. NMR spectroscopy

The detailed structure of EPS was studied using the NMR spectroscopy. The NMR spectroscopy of EPS was performed using JEOL ECZR series 600 MHz NMR spectrophotometer. The EPS sample (10 mg) was dissolved in D_2O (0.5 mL, 99.9%) solvent and ¹H, ¹³C, and 2D HSQC NMR studies were conducted. The operating frequency for ¹H- and ¹³C-NMR was 600 MHz and 150 MHz, respectively. The chemical shift (δ) for proton NMR and ¹³C-NMR was recorded in the range of 1–10 ppm and 10–200 ppm, respectively. ¹H-NMR spectra were acquired in total of 700 scan, whereas ¹³C-NMR spectra were obtained in total of 600 scans.

2.5.6. Determination of molecular weight

The weight-average molecular weight (Mw) and number average molecular weight (Mn) of EPS were analyzed by size exclusion chromatography. The GPC system was equipped with refractive index detector, PL aquagel-OH 40 column, and PLgel Mixed guard column. The filtered EPS sample was injected in the column and eluted at the flow rates of 1 mL/min at 35°C temperature. Dextran was used as a standard.

2.5.7. Field emission scanning electron microscopy and energydispersive X-ray spectroscopy (SEM-EDX)

The dried EPS sample was analyzed to study the surface morphology and microstructure using the field emission scanning electron microscope (FEG-SEM). The micrographs of the EPS surface were captured by operating the FE-SEM (JEOL-JAPAN; JSM-7600F) with an accelerated voltage of 10 kV. To study about the presence of various elements on the surface of the EPS, energy-dispersive X-ray spectroscopy was performed along with FE-SEM.

2.5.8. Atomic force microscopy (AFM)

Atomic force microscope (Asylum Research, USA, MFP-3D BIO) was used to study the molecular morphology of EPS. For the same, the polysaccharide was dissolved in distilled water at a concentration of 10 μ g/mL and was filtered through 0.45 μ m filter. Then, about 100 μ L of the polysaccharide solution was placed on freshly cleaved mica surface and dried at room temperature. Micrographs of EPS were taken using AFM under tapping mode.

2.5.9. Rheological behavior of EPS

The viscosity of EPS solution was measured using a Brookfield viscometer. The EPS was dissolved in distilled water to make 1% solution. The effect of temperature (in range of $30-70^{\circ}$ C) on the viscosity of the exopolysaccharide solution was measured at various shear rates (6–100 rpm). Moreover, the viscosity of EPS solution at various pH (1–13) was also measured. The viscosity of EPS solution was compared with guar gum.

3. RESULTS AND DISCUSSION

3.1. Purification of EPS

The dialyzed EPS was passed through a Sephadex-G75 column for further purification. As depicted in Figure 1, the EPS started eluting

from the 5^{th} fraction and the maximum elution was observed in the 20^{th} fraction. The EPS purification profiled is displayed in Table 1.

The EPS was purified 8.11-fold with 11.16% yield. In contrast to our study, Su *et al.* [9] reported purification of EPS from *Morchella aconica* using DEAE-Cellulose 52 anion-exchange column chromatography with 5% purification yield. However, Sephadex G-100 column chromatography was used to purify the EPS obtained from *Paecilomyces cicadae* [10]. Limin *et al.* [11] reported purification schizophyllan from *Schizophyllum commune* using Sephacryl S-500 column chromatography.

3.2. Characterization of EPS

3.2.1. Monosaccharide composition analysis

The detection of sugar monomers in EPS was performed using HPTLC [Figure 1a and b]. The compositional monosaccharides present in EPS were glucose and mannose. This confirms that the EPS is a glucomannan with heteropolymeric nature. Similarly, using TLC monosaccharide composition of EPS from *Lactobacillus fermentum* TDS030603 was reported to have two monomers such as glucose and galactose [12].

The HPLC chromatogram of the hydrolyzed EPS from *F. meliae* AGDP-2 demonstrates and confirms that glucose and mannose are the major

monomeric components present in varying concentrations [Figure 1c]. Herein, glucose and mannose were recorded as the predominant sugars showing peaks at retention times 10.170 and 10.995 min, respectively. EPS produced by a soil isolate *Streptomyces* sp. was also reported to be similarly composed of glucose and mannose monomers only [13]. Rani *et al.* [14] reported that probiotic *Bacillus tequilensis* FR9 isolated from the chicken gut was able to produce EPS which was composed of arabinose, glucose, galactose, mannose, and xylose as sugar monomers.

3.2.2. UV-visible spectroscopy

The UV–visible spectra of EPS in the range of 200–1000 nm were studied. The EPS showed no absorptions at 260 nm and 280 nm, indicating the absence of contaminants such as nucleic acids and proteins. Similar results were also reported by Yu *et al.* [15] who showed the absence of absorption at 260 and 280 nm for the UV scans of AMP-1 and AMP-2 polysaccharides extracted from *Armillaria mellea*.

3.2.3. FT-IR spectroscopy

The IR spectrum of EPS from *F. meliae* AGDP-2 displayed in Figure 2 shows a range of absorption peaks from 3410 to 555 cm⁻¹. The broad peak observed at 3410.99 cm⁻¹ is a characteristic presence



Figure 1: (a) Separation of hydrolyzed EPS on TLC sheet, (b) HPTLC chromatogram, and (c) HPLC chromatogram.

Table 1: Purification profile of EPS.

Samples	Volume (ml)	Total carbohydrate (mg)	Total protein (mg)	Carbohydrate/protein ratio	Yield (%)	Fold purification
Crude EPS	30	259.2	4.5	57.55	100	1
Dialyzed EPS	33	248.7	3.96	62.80	95.84	1.09
Sephadex G-75	33	28.95	0.062	466.93	11.16	8.11



Figure 2: FT-IR spectra of EPS.

of the stretching vibration of hydroxyl ion presence or hydroxyl groups (O-H). This confirms the polymeric nature of the sample and also indicates the strong intra- and inter-molecular interactions in EPS chains [16]. The absorption in this region also represents the carbohydrate rings in the structure which are responsible for water solubility of EPS [17]. A weak C-H stretching vibration near 2932 cm-1 was also observed. The C-N and the amide (>C=O) stretching vibration observed as prominent absorption peak at 1641.48 cm⁻¹, which indicated the characteristics of carbohydrates such as mannose [18]. The presence of carboxyl groups is indicated by a weak symmetrical stretching at 1373.45 cm⁻¹. The absorption in the region of 1000-1200 cm⁻¹ is observed due to the asymmetric stretch vibration of C-O-C and C-O-H in polysaccharides [19]. Finally, the absorption band at 828.30 cm⁻¹ indicates the presence of β -glycosidic linkage between glycosyl residues [20]. The diverse hydroxyl, amino, and carboxyl functional groups may serve as the binding sites for divalent cations and could contribute to the emulsification potential of EPS [21]. FTIR spectroscopy is also being used by researchers to study about functional groups and bonding patterns of polysaccharides [15,22,23].

3.2.4. Molecular weight determination

One of the most convenient methods to determine the molecular weight of polysaccharides is GPC. The weight-average molecular weight (Mw) of EPS produced of *F. meliae* AGDP-2 under submerged conditions was calculated to be 2.48 x 10⁶ Daltons and number average molecular weight (Mn) was estimated to be 2.19×10^6 Daltons. Molecular weight distribution, that is, polydispersity index of the EPS was 1.13. The similar method was used for estimating the molecular weight of polysaccharides produced by *S. commune* IBRC-M30213 and *Neopestalotiopsis* sp. strain SKE15 [23,24]. The weight-average molecular weights of these polysaccharides were found to be 37×10^4 Daltons and 1.5×10^6 Daltons, respectively. Molecular weights of these polysaccharides are quite less than that of EPS produced by *F. meliae* AGDP-2.

3.2.5. NMR analysis

The ¹H, ¹³C, and 2D HSQC NMR of EPS from *F. meliae* AGDP-2 demonstrated the complex and heterogeneous nature of EPS [Figure 3]. The ¹H-NMR spectrum of EPS contains numerous signals between δ 3.0 and 4.0 ppm that signifies ring proton region. Other signals

between δ 5.5 and 5.0 ppm represent the presence of α -anomers in the EPS structure [25]. The anomeric proton of α -D-mannose was indicated by a resonance at δ 5.1 ppm [26]. The signals recorded in the ring proton region δ 3.0–4.0 ppm could not be resolved properly due to very high molecular weight and overlapping chemical shifts and that can be described as key region of sugar molecules [27].

The ¹³C-NMR spectrum of EPS from *F. meliae* AGDP-2 consists of two major regions: (i) anomeric region (δ 90–100 ppm) and (ii) ring carbons regions (δ 60–80 ppm). The chemical shifts at δ 72.77, 75.77, and 75.96 ppm represent the presence of β -D-glucose in EPS structure [26]. Moreover, the prominent signal at δ 69.61 and 69.66 ppm was attributed to the presence of α -D-mannose residues. The presence of glucose components was confirmed by the downfield shift at δ 60.60 and 60.76 ppm which were allotted to a C-6 carbon signal and corroborated that the two glucose units in the polymer chain backbone were linked through α -(1 \rightarrow 6) linkages [28].

2D NMR analysis of EPS was also performed. The 2D HSQC NMR spectrum showed well-defined correlation in the ring proton regions. Goyzueta-Mamani *et al.* [29] also performed 2D HSQC NMR of chitin-like EPS produced by submerged fermentations of *Mortierella alpine.*

3.2.6. Surface morphology of EPS and elemental analysis

SEM is a powerful imaging technique to study and characterize the biological macromolecules and illustrating their common physical properties including surface morphology, microstructure, and texture. The SEM images [Figure 4a and b] revealed that the EPS from *F. meliae* AGDP-2 has a compact structure with crystalline structural moieties of an average size of 1.21 μ m. Moreover, some microcracks were also observed in the structure of EPS.

Similar methods were also employed by Vinothini *et al.* [22] for studying the surface of EPS from *Streptomyces griseorubens* GD5 where they observed compact and porous structure with flake-like structural units.

Energy-dispersive X-ray analysis technique has been widely utilized for quantitative elemental of a sample. In this study, seven elements including O, K, Na, Mg, S, Cl, and Ca were detected in the elemental analysis of the EPS from *F. meliae* AGDP-2 and their levels in terms



Figure 3: NMR spectra of EPS; (a) 1H NMR (b) 13C NMR, and (c) 2D HSQC NMR.

of weight% are also displayed [Figure 4c]. Such elements present on the EPS surface may have role in binding with carboxyl and hydroxyl groups of monosaccharides [14]. Similarly, during EDX analysis of EPS from *S. griseorubens* GD5, eight elements were detected [22].

3.2.7. AFM

AFM is a valuable technique to for topographical characterization. It can be employed to study the random linear or spherical structures of polysaccharides. EPS is found to be primarily comprised with spherical and irregular shaped lumps [Figure 5a and b]. The EPS formed large lumps which revealed about molecular aggregation, and the structures of polysaccharide molecules were branched and entangled, which can be attributed to their hydroxyl and carboxyl groups that can form strong intermolecular and intramolecular interactions with each other or with water molecules [30]. The average height of EPS ranged from 2 to 20 nm. According to Shen *et al.* [31] if mean height is more than 1 nm, then the sample may be branched and interwoven with one another.

The similar phenomenon is observed with the AFM analysis of the present study. Moreover, as the molecules are tightly packed,



Figure 4: (a and b) Scanning electron micrograph of EPS and (c) EDX analysis of EPS.



Figure 5: (a) Atomic force micrograph of EPS and (b) 3D visualization of AFM.

it can be said that EPS may have high affinity toward water and may possess pseudoplastic behavior [32]. Gu *et al.* [33] reported that polysaccharide SPU70-W1 from *Sagittaria sagittifolia* L. was mainly consisting of large spherical lumps and mean height was about 1.58 nm, which is very less than that is observed in our studies. Exopolysaccharide produced by *Streptococcus thermophilus* CC30 was also found to have lumps with mean height ranging from 10 to 30 nm [34].

3.2.8. Rheological behavior of EPS

The viscosity of EPS and guar gum at different shear rates (6-100 rpm) and temperatures $(30-70^{\circ}\text{C})$ was measured. It was observed that the viscosity remarkably decreased by increasing temperature and shear rate as depicted in Figure 6a and b. The viscosity of EPS at 30°C and 6 rpm was about 139 cP which

decreased to 68.8 cP at the shear rate of 100 rpm. However, viscosity of EPS was measured to be 58.5 cP at 70°C and 6 rpm that decreased to 44.57 cP at the shear rate of 100 rpm. The similar observations were also reported for guar gum. The results clearly indicated that increasing temperature and shear rate resulted in declining the viscosity of the EPS solution. This behavior indicates pseudoplastic property of the biopolymer. It is possible that the long chains of the EPS make an alignment in the direction of flow when the spindle speed increases. Therefore, fewer interactions among relative polymer chains are created. These findings were in agreement with those reported by Qiao *et al.* [35] for the biopolymer viscosity extracted from *Enteromorpha prolifera*.

The viscosity of 1% w/v aqueous solution of the EPS was assessed. As shown in Figure 6c, the viscosity of the EPS is dependent to changes in pH values. At the slightly acidic pH of 5, the viscosity of the EPS was 148 cP, while the viscosity declined in the alkaline environments. In contrast, the viscosity of guar gum was found to be maximum of about 294.4 cP at pH 9, and the same remained quiet low under acidic conditions. It is may be possible that shifting the pH value to either alkaline side may disturb the molecular conformation of EPS chains due to affecting by a net electric charge of the molecules [36].

4. CONCLUSION

The present study is focused on purification, characterization, and application of EPS produced by *F. meliae* AGDP-2. Using Sephadex-G75 column chromatography, EPS was purified by 8.11-



Figure 6: (a) Viscosity of EPS at various temperatures and shear rates, (b) VISCOSITY of guar gum at various temperatures and shear rates, and (c) viscosity of EPS and guar gum at different pH values.

fold. Furthermore, the chemical and structural analysis of F. meliae EPS was done by HPLC, HPTLC, UV-visible spectroscopy, FT-IR, GPC, NMR, SEM-EDX, and AFM. The monosaccharide compositional analysis revealed that EPS is glucomannan which is mainly composed of glucose and mannose monomers. The weightaverage molecular weight was calculated to be 2.48×10^6 Da. FTIR spectroscopy showed the occurrence of various functional groups and glycosidic linkage. NMR studies confirmed the presence of alpha-anomers of glucose and mannose residues. SEM analysis displayed micrometer sized crystal-like structures on EPS surface with some microcracks. Elemental analysis of EPS surface was done by EDX techniques which confirmed the presence of various elements including O, C, Na, Mg, S, Ca, and Cl. AFM studied revealed that EPS is composed of several spherical or irregular shaped lumps and is highly dense and branched. Moreover, studies on rheological behavior explained pseudoplastic behavior of EPS. Looking at the physicochemical properties, the purified EPS can be further investigated for the applications in various fields such as food industries, medical and pharmaceutical industries, as well as in white biotechnology.

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6. AUTHOR CONTRIBUTIONS

All authors made substantial contributions to conception and design, acquisition of data, or analysis and interpretation of data; took part in drafting the article or revising it critically for important intellectual content; agreed to submit to the current journal; gave final approval of the version to be published; and agree to be accountable for all aspects of the work. All the authors are eligible to be an author as per the international committee of medical journal editors (ICMJE) requirements/guidelines.

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8. CONFLICTS OF INTEREST

The authors declare that they have no known competing interest.

9. ETHICAL APPROVALS

This study does not involve experiments on animals or human subjects.

10. DATA AVAILABILITY

No additional or supplementary data files associated with the manuscript.

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