



Recent Advances in substrate utilization for fermentative hydrogen Production

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ABSTRACT

In many ways, hydrogen is considered to be the ideal fuel for the future. Biohydrogen production processes are considered as the most environmentally friendly in comparison to others. Biological hydrogen production processes (both dark- and photo- fermentation) are considered most favorable. Due to their high conversion efficiency and versatility of the substrate they can utilize. The use of waste materials as substrate not only generates energy but it also helps in the bioremediation. The potential utilization of waste material for H₂ production is being investigated extensively. The present review article aims to summarize the recently used substrates for fermentative biohydrogen production.

1. INTRODUCTION

Today, the world is facing two major issues; global climate change due to combustion of fossil fuels and energy crisis due to exhaustion of existing fossil fuels. Almost every country is concerned about this grave situation, and spending a lot of money to address the two issues. Fossil fuels are limited, and are on the verge of their depletion. The depletion of fossil fuels raises the question to find the alternative energy sources, and ranks as one of the most challenging problems of mankind. H₂ is considered as a clean energy with no CO₂ emissions having high energy yield of 122 KJ/g, 2.75 times over the hydrocarbon fuels [1-2]. H₂ seems a promising candidate to replace the fossil fuels, and is produced biologically via biophotolysis, dark-fermentation and photo- fermentation. Hydrogen produced biologically is known as biohydrogen. The advances in biohydrogen production technologies based on organic wastewater conversions could solve the issues pertaining to food security, climate change, energy security and clean development in the near future [3]. Biological methods of H₂ production

are preferred over the chemical ones, as they are less energy intensive, and utilize organic wastes as the feedstock. Biohydrogen could be produced commercially at large scale by using efficient H₂ producing microorganisms from the readily available, renewable substrates. The cost of the substrate plays a major role in the economics of biohydrogen production. Therefore, scientists working in the area of biohydrogen have more concern for the cost of substrate.

Cheaper and abundant feedstock can make the biohydrogen process economically viable. The efficient utilization of organic wastes from industries and agriculture as substrate for biohydrogen production not only supports green energy generation but also helps in bioremediation [4]. Pure sugars are also used as substrate for biohydrogen production, but they are costly. Wastes like agricultural residues [5], food wastes [6] and effluents from industrial processes such as dairy wastewater [7] olive processing [8] and cheese production [9] can be used as substrate for biohydrogen production.

Utilization of these wastes as substrate for biohydrogen proves to have dual economic benefit of energy production and savings in the cost of waste disposal. The present review is an attempt to narrate recent advances in substrate utilization for biohydrogen production, and specific attention has been paid to the recently used substrates for sustained H₂ production on a cost-effective basis.

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2. PROCESSES FOR BIOHYDROGEN PRODUCTION

Biologically hydrogen can be produced by direct biophotolysis, indirect biophotolysis, dark-fermentation and photo-fermentation [3]. Direct-biophotolysis is carried out in the presence of sunlight by green algae using readily available water as substrate.

In direct biophotolysis light energy absorbed by PSII and PSI helps to transport electrons from water to ferredoxin and reduced ferredoxin acts as an electron donor to a hydrogenase enzyme, which reversibly catalyzes the reduction of proton (H^+) to molecular hydrogen [10]. The main disadvantage of this process is low light conversion efficiencies and O_2 labile hydrogen production system. Indirect biophotolysis is carried out by filamentous cyanobacteria both nitrogen fixing and non-nitrogen fixing.

In cyanobacteria photosynthesis and nitrogen fixation reactions are spatially separated from each other. Major drawbacks of this process are presence of uptake hydrogenase enzyme and low H_2 production rates. Dark-fermentation carried out under anaerobic conditions seems to be more favourable, since hydrogen is yielded at a high rate with various organic substrates and waste waters enriched with carbohydrates [11]. Currently, dark-fermentation technologies are under development at laboratory scale to produce biohydrogen from organic wastes. During the dark-fermentation simple sugars are converted into H_2 , VFAs and alcohols which are organic pollutant and energy carriers. In photo-fermentation process, under anaerobic conditions, photosynthetic bacteria use light as energy source and assimilate small organic acids to produce biomass, H_2 and CO_2 . Purple non-sulfur bacteria (PNS) are considered as promising candidates for photo-fermentative H_2 production due to their ability of high substrate conversion efficiencies, being able to utilize wide range of the solar spectrum and flexibility in utilization of wide variety of organic wastes [1]. The use of dark-fermentation along with photo-fermentation as second step not only allows the effective utilization of the substrate but also improve the overall H_2 yield [12].

3. SUBSTRATES FOR BIOHYDROGEN PRODUCTION

The use of synthetic media for microbial growth and maintenance is ideal; however, their use for commercial scale is not suitable because of the cost involved. Hence, for the commercial production of biohydrogen, utilization of various domestic, agricultural and industrial wastes rich in organic matter not only lowers the production costs, it also helps in removal of these wastes from the environment [12].

However, there is a need of pre-treatment of these waste materials before their use for biohydrogen production. The pre-treatment of the wastes potentially enhances the biodegradability of the wastes by microorganisms thus increasing the efficiency/yield of biohydrogen production. The various types of wastes used for biohydrogen production are:

3.1 Industrial waste

Considerable research is going on for the utilization of industry waste as substrate for biohydrogen production (Table 1). The food industry produces wastes/wastewaters highly concentrated in carbohydrates e.g. sugars, starch and cellulose. The high organic loading and other growth factors that support microbial growth makes such wastes/wastewaters the potential and important feedstock for biohydrogen production. Before using them as substrate for biohydrogen production, such wastewaters need to be pre-treated for pH and nutrient balance. Many wastes or wastewaters like olive mill effluent consist of ethanol, acetate, butyrate, and propionate, and are reported to be best suited for photo-biological H_2 production. Kim *et al.*, [13] investigated the potential of using the industrial effluents of Makkoli (raw rice wine), Tofu (soybean curd) wastewaters and sewage sludge as substrates for H_2 production in combination with dark-fermentation by *C. butyricum* NCIB 9576 and photo-fermentation of the spent media by purple non-sulfur photosynthetic bacteria. From the Makkoli wastewater, approximately 1L H_2 /L wastewater was produced during the dark-fermentation. About 0.44 L H_2 /L broth /day was produced during photo fermentation. Tofu wastewater, generates about 0.9 L H_2 /L wastewater during dark- and 0.2 L H_2 /L broth /day during photo-fermentation. Sewage sludge was pre-treated for 1h at 150 °C under 10 atm after alkali treatment before it could be used as substrate. It continuously produced 0.17-0.28 1L H_2 /L broth/day during photo-fermentation. The data suggested that organic rich industrial effluents of Makkoli and Tofu wastewater and sewage sludge were the promising substrates for H_2 production and for the treatment of organic waste and wastewater. Eroglu *et al.* [8] used Olive mill wastewater (OMW) containing carbon, hydrogen, and nitrogen in suspended solids as sole substrate for the biohydrogen production by *Rhodobacter sphaeroides* O.U.001 in glass column-photobioreactors. OMW was diluted in the range of 20% (v/v) and 1% (v/v) OMW containing media.

Maximum hydrogen production potential (HPP) was found at 2% OMW. During the fermentation process, chemical oxygen demand (COD) of the diluted wastewater decreased from 1100 to 720 mg/L; biochemical oxygen demand (BOD) decreased from 475 to 200 mg/L. The results concluded OMW to be a very promising substrate for biohydrogen production. Seifert *et al.* [7] used dairy wastewater as substrate for hydrogen production by *R. sphaeroides* O.U. 001. Concentration of the waste varied from 5 to 40 %, v/v, keeping the inoculum 0.36 g dry wt/L under illumination of 9 klx. The highest volumetric hydrogen was obtained at concentration of 40 v/v %, but the maximal substrate yield was at lower concentrations of the waste (5–10 v/v %). Cappelletti *et al.* [14] used cassava processing wastewater for biohydrogen production by *Clostridium acetobutylicum* ATCC 824.

They studied the effect of initial substrate concentration on COD consumption, pH, and H_2 production. Higher substrate concentrations (30.0 and 15.0 COD/L) led to lower H_2 yield and substrate conversion efficiency.

Table 1: Industrial waste/wastewaters used as substrate for biohydrogen production.

S. No.	Waste/waste water	Inoculum	Highest H ₂ yield	Reference
1.	Cheese whey	<i>Clostridium saccharoperbutylacetonicum</i>	0.028 L/h	[60]
2.	Cheese processing waste water	Sewage sludge	2.3 mmol/g COD	[61]
3.	Cheese whey waste water	Anaerobic digester sludge	22 mmol H ₂ /g COD	[62]
4.	Cereal waste water	Dewatered sewage sludge	0.79 mol H ₂ /mol glucose	[63]
5.	Cheese whey	Anaerobic granular sludge	2.8 mol H ₂ /mol hexose	[64]
6.	Cheese whey	<i>E. aerogenes</i>	2.04 mol H ₂ /mol lactose	[9]
7.	Cassava waste water	Pond sludge	1.91 mol H ₂ /mol glucose	[65]
8.	Olive mill waste water	<i>Rhodobacter sphaeroides</i> O.U.001	13.9 L H ₂ /L OMW	[8]
9.	Cassava processing waste water	<i>Clostridium acetobutylicum</i> ATCC 824	2.41 mol H ₂ /mol glucose	[14]
10.	Dairy waste water	<i>Rhodobacter sphaeroides</i> O.U.001	3.6 L H ₂ /L dairy wastewater	[7]

Whereas, low COD concentrations increased H₂ yield up to 2.41 mol H₂/mol glucose, with efficiency of 60% (mol/mol), respectively. The results demonstrated the successful utilization of cassava processing wastewater for H₂ production by *C. acetobutylicum*. Cheese whey, the by-product of cheese industry has high organic load. The presence of high amount of lactose and minerals in whey make it a potential substrate for biohydrogen production [15-17]. Rai *et. al.* [9] reported biohydrogen production from cheese whey in a two- step dark- and photo-fermentation process by free and immobilized bacterial cells. The cumulative H₂ yield for free and immobilized bacterial cells during two- step process was 3.40 and 5.88 mol/mol lactose, respectively. The data suggested that the use of cheese whey as substrate with immobilized bacterial cells has good potential for biohydrogen production and effective removal of organic load from the wastewater. Further, Rai *et. al.* [17] optimized the concentration of essential trace elements required for growth/metabolism of most microorganisms to enhance the hydrogen yield during photo-fermentation step. Spent medium generated after dark-fermentation of cheese whey wastewater by *Enterobacter aerogenes* MTCC 2822, was subjected to photo-fermentation through enrichment by Ni²⁺ (0-8 µmol/L), Fe²⁺ (0-100 µmol/L) or Mg²⁺ (0-15 mmol/L) by *Rhodospseudomonas* BHU 01 strain. The results clearly indicated 4 µmol Ni²⁺/L. Fe²⁺ (60 µmol/L) resulted in maximum cumulative H₂ production and yield. Nevertheless, even 6 mmol of Mg²⁺ did not significantly affect H₂ production (110 ml) or yield (44 mmol). The observations suggested the role of Fe²⁺ and Ni²⁺ in regulation of nitrogenase and hydrogenase enzyme, while Mg²⁺ mainly in the biosynthesis of photopigment bacteriochlorophyll (Bchl). Xiao *et. al.* [18] reported bio-hydrogen production from protein wastewater by altering protein structure and amino acids acidification type via pH control. The hydrogen production reached 205.2 mL/g protein when protein wastewater was pre-treated at pH 12 and then fermented at pH 10. The studies showed that the pre-treatment significantly enhanced protein bio-hydrolysis during the subsequent fermentation stage as it caused the unfolding of protein, damaged the protein hydrogen bonding networks, and destroyed the disulfide bridges, which increased the susceptibility of protein to protease.

3.2. Food waste

Food waste rich in sugars is an attractive substrate for biohydrogen production (Table 2). The presence of moisture

(72-85.2%), high organic load (COD: 19.3-346 g/L; carbohydrate: 25.5-143 g/L) and high carbon to nitrogen (C/N) ratio (9-21) make food waste a suitable substrate for biohydrogen production [19-21]. In addition, other characteristics such as volatile solid composition, particle size and biodegradability are also important to achieve high hydrogen yield [22]. Food composition varies from source to source thus influencing the process parameters drastically.

High concentration of proteins, fats and lipids in the food waste adversely affects the hydrogen production as these are not easily degraded by the microorganisms employed for hydrogen production. The presence of lipids and fats in an anaerobic fermenter resulted in flotation, clogging and mass transfer problems. Use of high temperature is suggested to prevent the clogging problem during anaerobic degradation of lipids [23]. But increasing the temperature beyond the optimum range of the fermentation process decreases the hydrogen yield. Pre-treatment of food waste is regarded as the important parameter in influencing biohydrogen production and yield [24]. Pre-treatment of food waste is done to eliminate the indigenous microflora present in the food waste. The indigenous mixed microflora may contain hydrogen producing or hydrogen consuming bacteria, methane producing bacteria and acid producing bacteria. Therefore, pre-treatment of the indigenous microorganisms present in food waste by heat, chemical or pH shock to promote growth of hydrogen-producing bacteria and elimination of hydrogen-consuming bacteria [25]. Many workers reported the dominance of hydrogen producing bacteria such as *Clostridium* sp., and *Caloramator australicus* after heat treatment for biohydrogen production study using food waste as the substrate [26-28]. In case, no pre-treatment is applied to the food waste microflora before its use as substrate the number of methane- and acid-producing bacteria will increase and hydrogen-producing bacteria remain encapsulated by the spores. The heat pre-treatment is chosen by most because of its simplicity in application, short time and low cost. On the other hand, pre-treatment by chemicals (alkali, acid or chemical shock) takes long time, and is costly [25, 29].

A combination of pre-treatments is also reported by Elbeshbishy *et. al.* [30]. Four individual pre-treatment methods: ultrasonication, heat, acid, and base and three combined pre-treatment methods comprising ultrasonication with heat (UH), ultrasonication with acid (UA), and ultrasonication with base (UB) were applied on the food waste.

Table 2: Food waste used as substrate for biohydrogen production.

S. No.	Food waste	Inoculum	Highest H ₂ yield	Reference
1.	Food waste	Heat shock treated anaerobic sludge	310 mL H ₂ /g VS _{added}	[66]
2.	Food waste	Thermophilic acidogenic culture	46.3 mL H ₂ /g VS _{added}	[31]
3.	Rice slurry	Anaerobic digested sludge	346 mL H ₂ /g carbohydrate	[6]
4.	Pine apple waste	Municipal sewage sludge	5920 mmol H ₂ /g COD	[67]
5.	Kitchen waste	Digested slurry	72 mL H ₂ /g VS _{added}	[19]
6.	Wasted bread	Rice rhizosphere	1.3 mol H ₂ /mol hexose	[68]
7.	Vegetable kitchen waste	Kitchen waste compost	38 mL H ₂ /g COD	[23]
8.	Fully ripened fruits	Sewage sludge	2.2 mol H ₂ /mol glucose	[20]
9.	Kitchen waste	Mixed sludge	96 mL H ₂ /g VSS	[69]
10.	Food waste (rice, fish and vegetable)	Palm oil mill effluent sludge	79 mmol H ₂ /L media/day	[28]

There is a lot of variation in characteristics of food waste being used for biohydrogen production as the type of waste generated in different countries is different. Zhang *et al.* [22] collect food waste in the city of San Francisco, California, USA and characterized its potential as the feedstock for anaerobic digestion processes. They measured the daily and weekly variations of food waste composition over a two-month period due to daily variation in the food waste composition. Water is generally added to homogenize/dilute the food waste before it is being used [31-32]. The dilution of food waste facilitates the use of optimum concentration of sugars (COD) for fermentative hydrogen production. Hwang *et al.* [20] used different ripened fruits (apple, pear, and grape) as feedstock for hydrogen production in two-stage fermentation. Ripened apple was the most promising substrate for cumulative H₂ production with a maximum H₂ yield (2.2 mol H₂ /mol glucose) in the first stage, and additional cumulative biohydrogen (3337.4 mL H₂/L culture) in the second stage. The study demonstrated that ripened fruits could be used as substrates for biohydrogen fermentation. A novel process that produces H₂ without inoculum addition was reported by Kim *et al.* [24]. Food waste utilized as substrate acts not only as a substrate but also as a source of H₂-producing microflora in case heat (90 °C for 20 min), acid (pH 1.0 for 1 d), or alkali (pH 13.0 for 1 d) treatment was applied. The effect of initial pH on hydrogen production from food waste was investigated by Kim *et al.* [33]. At initial pH 8.0 H₂ yield 1.3-1.9 mol H₂/mol hexose_{added} was achieved which corresponded to 8.13% of the total energy content in the substrate. Vegetable based market waste was evaluated as the substrate for hydrogen production using selectively enriched acidogenic mixed consortia under acidophilic microenvironment by Venkata Mohan *et al.* [34]. Different substrate/organic loading conditions in concurrence with two types of feed compositions (with and without pulp) were investigated. H₂ production was found to be dependent on the substrate concentration and composition, and it was high in experiments performed without pulp. Results indicated the feasibility of vegetable waste for H₂ production. A pilot-scale anaerobic sequencing batch reactor (ASBR) treating food waste was used for H₂-production by Kim *et al.* [35]. To vary the carbon/nitrogen (C/N) ratio from 10 to 30, the composition of the food waste was changed. It was found that when the C/N ratio was lower than 20, the H₂ yield was maintained at around 0.5 mol H₂/mol hexose_{added}, but it gradually dropped at higher C/N ratios mainly due to the

production of lactate, propionate, and valerate. Alkaline shock (pH 12.5 for 1 day) was given to recover the yield, and it was so effective that the H₂ yield significantly increased to over 0.9 mol H₂/mol hexose_{added}, and then got stabilized at 0.69 mol H₂/mol hexose_{added}. Canteen based composite food waste was used as substrate for biohydrogen production in an anaerobic sequencing batch reactor (AnSBR) at pH 6 with five variable organic loading conditions (OLR1, 0.854; OLR2, 1.69; OLR3, 3.38; OLR4, 6.54 and OLR5, 9.85 kg COD/m³/day) to find the influence of carbohydrates and proteins concentration on fermentative hydrogen production [36]. The data suggested that H₂ production depends on the substrate load. OLR4 supported maximum H₂ production (69.95 mmol), while higher substrate degradation (3.99 kg COD/m³/day) was observed with OLR5.

3.3. Lignocellulosic biomass

Lignocellulosic biomass is considered as the most abundant raw material found in nature ranging from hardwood, soft wood, grasses, agricultural and forestry residues as well as secondary biofuels wastes. The annual worldwide yield of lignocellulosic biomass residue is estimated to exceed 220 billion tons [37]. Although, being an abundant and almost zero cost feedstock, agricultural and forestry residues do not contain easily fermentable free sugars, but complex carbohydrate polymers, i.e., cellulose and hemicelluloses, which are tightly bonded to lignin [38]. Thus utilization of these as a substrate for biohydrogen production is not an easy task due to the presence of non biodegradable lignin coating. Therefore, prior to use biomass as a substrate, pre-treatments such as physical, chemical and biological or a combination of these seem necessary (Table 3). The pre-treatment of the biomass decreases the crystallinity and increase the surface area to improve the consumption by the microorganisms and yield simple sugars for fermentation. Cellulose rich corncobs are abundant in nature and its composition makes it a potential substrate for H₂ production. Hydrogen production from acid and enzyme pre-treated corncob by integrating dark-fermentation with photo-fermentation was investigated by Yang *et al.* [39]. In the first step, dark fermenting microflora obtained from dairy manure produced maximum biohydrogen yield and rate 120.3 ± 5.2 mL H₂/g corncob and 150 mL H₂/L h, respectively. In the second step, *Rhodobacter sphaeroides* isolated from sedimentation tank of sewage plant yielded 713.6 ± 44.1 mL H₂/g-COD hydrogen.

Table 3: Lignocellulosic waste used as substrate for biohydrogen production.

S. No.	Biomass	Inoculum	Highest H ₂ yield	Reference
1.	Sugarcane bagasse	<i>C. butyricum</i>	1.73 mol H ₂ /mol total sugar	[70]
2.	Sweet sorghum bagasse	<i>C. saccharolyticus</i>	73.6 mL H ₂ /mmol hexose sugars	[71]
3.	Cornstalk	Heat pre-treated anaerobic sludge	126.22 mL H ₂ /g dry biomass	[72]
4.	Wheat Straw	Anaerobic mixed culture	22.4 mL H ₂ /mmol glucose	[73]
5.	Lawn grass	Enriched mixed culture dominated by <i>C. pasteurianum</i>	72.21 mL H ₂ /g dry biomass	[74]
6.	Soybean straw	Enriched mixed culture dominated by <i>C. butyricum</i>	60.2 mL H ₂ /g dry biomass	[75]
7.	Water hyacinth and beverage wastewater	Pig slurry	13.65 mL H ₂ /g dry feedstock	[76]
8.	Corn stalk	<i>T. thermosaccharolyticum</i> W 16	89.3 mL H ₂ /g dry biomass	[77]
9.	Sugarcane bagasse	<i>E. aerogenes</i>	1000 mL/L hydrolysate	[44]
10.	Corn stalk	<i>C. butyricum</i>	92.9 mL H ₂ /g dry biomass	[78]

This study successfully demonstrated the feasibility of converting corncob into high-yield hydrogen. Cassava is a low cost starch rich crop used for production of starch. Cassava pulp is a leftover residue from starch production industry, which contains as high as 50- 60 % starch content on dry basis [40], while the other main carbohydrates are cellulose and hemicelluloses [41]. To utilize the cellulose and hemicellulose in cassava pulp, it was acid hydrolysed and used as substrate for fermentative hydrogen production by anaerobic mixed cultures by Phowan and Danvirutai [42]. The optimal conditions for hydrolysis found were 0.5% H₂SO₄ at the ratio of 1:15 (dry wt.: volume) for 30 min and yielded 27.4 g/L of total sugar. The experiments were carried out to determine the optimal operating conditions for hydrogen production. The results indicated that the Up-flow Anaerobic Sludge Blanket (UASB) granules yielded the highest hydrogen production at the initial total sugar concentration of 25 g COD/L with the biomass concentration of 3.0 g/L and the initial medium pH of 5.5. The highest hydrogen yield of 342 ml H₂/g COD_{reduced} and the hydrogen production rate of 3381 ml H₂/L/d was obtained. The study demonstrates the possibility of using cassava pulp acid hydrolysate as the substrate for hydrogen production by the selected mixed culture. Cheng *et. al.* [43] used mixed bacteria in a two-step process to improve H₂ yield from cassava starch. In dark-fermentation, the mixed H₂ producing culture (having more *Clostridium* sp.), was used in batch mode. Substrate concentration, fermentation temperature and pH were optimized as 10.4 g/L, 31 °C and 6.3 by response surface methodology (RSM). The maximum hydrogen yield and production rate in dark fermentation were 351 ml H₂/g starch (2.53 mol H₂/mol hexose) and 334.8 ml H₂/L/h, respectively. For photo-fermentation, mixed immobilized photosynthetic bacteria (mainly *R. palustris*) were used. The maximum hydrogen yield in photo fermentation was 489 ml H₂/g starch (3.54 mol H₂/mol hexose). The total hydrogen yield was significantly increased from 402 to 840 ml H₂/g starch (from 2.91 to 6.07 mol H₂/mol hexose) by mixed bacteria and cell immobilization in combination of dark and photo fermentation. Rai *et. al.* [44] used acid hydrolysed sugarcane bagasse (SCB) as substrate for biohydrogen production by integrating dark- and photo-fermentation. The SCB was hydrolysed by sulphuric acid and detoxified to remove the inhibitory furfural, and subjected to dark-fermentation by *Enterobacter aerogenes* MTCC 2822. Photo-fermentation of the spent medium was done by *Rhodospseudomonas* BHU 01. The acid hydrolysis residue was

hydrolysed by *Cellulomonas fimi* to release sugars for H₂ production by *E. aerogenes*, through simultaneous saccharification, filtration and fermentation (SSFF). Cumulative H₂ production during dark-fermentation and SSFF was 1000 and 613 ml/L, respectively. During photo-fermentation of spent media of dark-fermentation and SSFF by *Rhodospseudomonas* BHU 01 the cumulative H₂ production was 755 ml/L and 351 ml/L, respectively. Kumar *et. al.* [45] used hydrochloric acid- pre-treated hydrolysate of de-oiled jatropha waste (DJW) as substrate for biohydrogen production in batch mode. The optimal hydrolysate concentration, temperature and pH were found as 10.2 g reducing sugar (RS)/L, 37 °C and 5.0, respectively. The optimum values were used subsequently to run CSTR (continuously stirred tank reactor) at various hydraulic retention times (HRTs). Maximum hydrogen production rates (HPR) were reported to be 0.86 L H₂ /L/d and 0.15 L H₂/L/d from batch and CSTR operations, respectively. Hydrochloric acid pre-treatment enlarged the pore size of the unhydrolyzed biomass (UHB) from 0.6 to 3.9 mm³/g. Kumar *et. al.* [46] reported hydrogen production from cello-lignin fraction of de-oiled jatropha waste (DJW) as substrate for biohydrogen production using hybrid immobilized cells. The cello-lignin fraction used was the non-hydrolyzed residue of acid pre-treated DJW. For releasing the reducing sugars from non hydrolyzed fraction, enzyme pre-treatment was given. Hydrogen production was seen in a CSTR at various HRTs. Results showed a peak HPR of 3.65 L H₂/L/d and HY of 150 mL H₂/g reducing sugars at the optimum HRT of 12 h and 37 °C. Use of energy crops as feedstock for energy production has been studied for long time. But the use of energy crops as substrate for biohydrogen production can only be justified if the production cost of crops is low, has high biomass and is rich in fermentable sugars. However, the food vs. fuel debate laid down the use of energy crops as source of energy as there is a competition for land between food and energy crops.

3.4. Algal biomass

Use of algal biomass as substrate for biohydrogen production seems the most promising step towards sustainable energy generation. Algal *spp.* have relative high photosynthetic efficiencies, high growth rates, and can thrive in various water sources ranging from brackish water to wastewater from food- and agro-industrial sector. Algal biomass can be utilized by H₂ producing microorganisms similar to other organic wastes as it is

rich in carbohydrates, proteins and lipids (Table 4). It has added advantage that it can fix CO₂ and unlike energy crops, it does not create 'land for food or for energy issue'. The complex carbohydrates present in algal biomass, are bound with rigid algal cell walls [47-48]. For utilization of these carbohydrates for fermentation, it is necessary to break the algal cell wall along with complex carbohydrate to facilitate the release of simple sugar [49]. For breaking the cell wall of algae and hydrolyzing the complex carbohydrates, several pre-treatments are employed such as physical (sonication, milling, grinding), chemical (acid, alkali) and biological methods (enzymatic). Each and every pre-treatment method has its own merits and demerits. Physical methods are simple but energy intensive, biological ones are costly and time consuming, so chemical methods are preferred over others because of higher conversion efficiency of complex carbohydrates into simpler fermentable sugars in lesser time [48, 50]. Pre-treatment increases the production cost of biohydrogen thus affects the overall economics of biohydrogen production from algal biomass. Kawaguchi *et. al.* [47] used marine green alga *Dunaliella tertiolecta* and the fresh water green alga *Chlamydomonas reinhardtii* biomass for biohydrogen production by a mixed culture of *Lactobacillus amylovorus* and *Rhodobium marinum* A-501. *L. amylovorus* utilized algal starch for lactic acid production and *R. marinum* A-501 produced hydrogen in the presence of light using lactic acid as an electron donor. The yields of hydrogen obtained from starch contained in *D. tertiolecta* and *C. reinhardtii* were 61% and 52 %, respectively, in the mixed cultures of *L. amylovorus* and *R. marinum* A-501. Nguyen *et. al.* [51] utilized the accumulated starch in the green algae *C. reinhardtii* as substrate for H₂ production by the hyperthermophilic eubacterium *Thermotoga neapolitana*. The bacterium having amylase activity could directly ferment the algal starch into H₂ to the extent of 1.8-2.2 mol H₂/mol glucose. To enhance the hydrogen production two pre-treatment methods using the heat- HCl treatment and enzymatic hydrolysis were applied on algal biomass before using it as substrate for H₂ production. The use of starch pre-treated by 1.5% HCl at 121°C for 20 min resulted in H₂ yield of 58% (v/v) whereas the enzymatic digestion of starch by thermostable alpha

amylase applied in the SHF process significantly enhanced H₂ productivity of the bacterium to 64% (v/v) of total accumulated H₂ level and a hydrogen yield of 2.5mol H₂/mol glucose. Results demonstrated that direct H₂ production from algal biomass was more cost effective as only one bacterial cultivation step was required for H₂ production. Park *et. al.* [52] investigated the feasibility of hydrogen production from red algae biomass. Galactose, the main sugar of red algae, was successfully utilized for H₂ production. The maximum H₂ production rate and yield from galactose were 2.46 L H₂/gVSS/d and 2.03 mol H₂/mol galactose_{added} respectively, which were higher than those for glucose (0.914L H₂/gVSS/d and 1.48 H₂/mol galactose_{added}). The main byproduct of acid hydrolysis, 5- hydroxyl methyl furfural (HMF) caused inhibition in H₂ fermentation and decreased H₂ production rate by 50% compared to the control. But when red algae biomass was hydrolyzed at 150 °C (15 min) and detoxified by activated carbon, 53.5 ml of hydrogen was produced from 1 g of dry algae with a hydrogen production rate of 0.518 L H₂/gVSS/d. They suggested that red algae could be the suitable substrate for H₂ production. Jung *et. al.* [53] tested marine algae for their use as a substrate in fermentative H₂ production by seed sludge microflora obtained from an aerobic digester in a local wastewater treatment plant. Among the algae tested, *Laminaria japonica* exhibited highest H₂ yield of 69.1 ml H₂/g COD added, and this was attributed to its high carbohydrate content and main constituent of alpha polysaccharides laminarine and alginates. To enhance H₂ production from algal biomass, thermal pre-treatment was applied under various conditions. At 170 °C (20 min), H₂ yield was maximum (109.6 ml H₂/g COD_{added}). Jung *et. al.* [54] applied a combined (acid + thermal) pre-treatment for enhanced fermentative H₂ production from *L. japonica*. Various pretreatment conditions including HCl concentrations, heating temperatures and reaction times were optimized via response surface methodology (RSM) with a Box-Behnken design (BBD). The desirable pre-treatment conditions found were HCl concentrations 4.8%, temp 93°C and reaction time 23 min, under which H₂ yield reached 159.6mL H₂/g dry cell weight. The main organic acids produced were acetic and butyric acid. The HMF, a byproduct formed

Table 4: Algal biomass used as substrate for biohydrogen production.

S. No.	Biomass	Inoculum	Highest H ₂ yield	Reference
1.	<i>Chlamydomonas reinhardtii</i>	<i>Thermotoga neapolitana</i>	311.1 mL H ₂ /g monosaccharides	[51]
2.	<i>Laminaria japonica</i>	Anaerobic mixed culture	71.4 mL H ₂ /g dry biomass	[79]
3.	Microalgae	Enriched functional consortia	25.1 mL H ₂ / g dry biomass powder	[80]
4.	<i>Chlorella vulgaris</i>	Heat pretreated anaerobic digestion sludge	33.8 mL H ₂ /g VS	[81]
5.	<i>Anabaena spp.</i>	<i>Enterobacter aerogenes</i>	15.1 mL H ₂ /g VS	[82]
6.	<i>Arthrospira platensis</i>	Heat pretreated anaerobic digestion sludge	101.7 mL H ₂ /g VS	[83]
7.	<i>Gelidium amansii</i>	Heat pretreated anaerobic digestion sludge	44.6 mL H ₂ /g VS	[56]
8.	<i>Nannochloropsis oceanica</i>	Heat pretreated anaerobic digestion sludge	39.9 mL H ₂ /g VS	[84]
9.	<i>Laminaria japonica</i>	Anaerobic digester sludge	61.3 mL H ₂ /g dry biomass	[85]
10.	<i>Chlorella sorokiniana</i>	<i>Enterobacter cloacae</i>	201.6 mL H ₂ /g COD	[46]
11.	Untreated de-oiled algae cake	Anaerobic digester sludge	66 mL H ₂ /g algal mass	[59]
12.	<i>Scenedesmus obliquus</i>	<i>Enterobacter aerogenes</i>	57.6 mL H ₂ /g VS	[86]
13.	<i>Scenedesmus obliquus</i>	<i>Clostridium butyricum</i>	113.1 mL H ₂ /g VS	[86]

during the pre-treatment process, showed an inverse relationship with H₂ yield, indicating that pre-treatment conditions for H₂ production from *L. japonica* were successfully optimized by increasing the hydrolysis rate of the feedstock and also reducing the formation of HMF. Liu *et. al.* [55] utilized carbohydrate-rich microalgal biomass of *Chlorella vulgaris* ESP6 for biohydrogen production. Algal biomass was hydrolyzed by acid or alkaline/enzymatic treatment, and was then used as substrate for biohydrogen production by *C. butyricum* CGS5. The biomass of *C. vulgaris* ESP6 was efficiently hydrolyzed by acid treatment with 1.5% HCl, giving a reducing sugars (RS) yield of nearly 100%. The optimal conditions for hydrogen production were 37 °C and the microalgal hydrolysate loading of 9 g RS/L with pH-controlled at 5.5. Under optimal conditions, cumulative H₂ production, production rate, and yield were 1476 ml/L, 246 ml/L/h, and 1.15 mol/mol RS, respectively. The results demonstrated that the *C. vulgaris* biomass has the potential to serve as effective feedstock for dark fermentative H₂ production. Kumar *et. al.* [48] investigated biohydrogen production from algal biomass of *Chlorella sorokiniana* as substrate by *Enterobacter cloacae* IIT-BT-08. The pre-treated algal biomass (10 g/L) with 2% (v/v) HCl-Heat was found most suitable for hydrogen production yielding 9±2 mol H₂/kg COD reduced, and was fitting with Gompertz equation. The use of green algae as substrate for H₂ production also helps in CO₂ sequestration from the environment as green algae have the capacity to absorb CO₂ from flue gas and eliminate the impact of global warming due to increasing concentration of CO₂ in the atmosphere. Park *et. el.* [56] used marine algal biomass of *Gelidium amansii* and powdered *G. amansii* was hydrolyzed at (120-180 °C), solid/liquid (S/L) ratio of 5-15% (w/v), and H₂SO₄ conc of 0.5-1.5% (w/w), and then fed to batch hydrogen fermentation using seed sludge obtained from anaerobic digestion in the local wastewater treatment plant. The maximum hydrogen production of 0.51 L H₂/L/h and 37.0mL H₂/g dry biomass was found at 161-164°C hydrolysis temp, 12.7-14.1% S/L ratio and 0.50% H₂SO₄. Roy *et. al.* [49] used *C. sorokiniana* pre-treated algal biomass as substrate for thermophilic biohydrogen production using mixed culture. The *C. sorokiniana* was cultivated in helical airlift photobioreactor and the resulting biomass was subjected to various physical and chemical pre-treatments. It was observed that pre-treatment with 200 dm³/m³ HCl heat was the most suitable pre-treatment resulting in cumulative H₂ of 1.93 m³/m³ and H₂ yield of 958 dm³/kg volatile suspended solid or 2.68 mol/mol of hexose. Liu and Wang [57] reported hydrogen production from algal biomass of *L. japonica* by anaerobic mixed bacteria. They investigated saccharification efficiency and hydrogen production by *L. japonica* with four different pre-treatment methods, including heat, acid, alkaline and ultrasonic treatment. The results showed that saccharification efficiency from *L. japonica* pre-treated with acid was the highest among the four methods. The maximum hydrogen production (83.45 ±6.96 mL/g) was achieved with *L. japonica* pre-treated with heat and an initial pH and substrate concentration of 6.0 and 2%, respectively. They suggested that the marine macro-alga *L. Japonica* could be a

utilized as a substrate for fermentative hydrogen production. Nayak *et. al.* [58] utilized *Anabaena* PCC 7120 biomass as substrate for biohydrogen production by thermophilic dark fermentative hydrogen production using mixed microflora. Maximum H₂ production was found to be 1600 mL/L upon pre-treatment with amylase followed by thermophilic fermentation (24 h) compared to other methods like sonication (200 mL/L), autoclave (600 mL/L) and HCl treatment (1230 mL/L). Amylase pre-treatment yielded higher reducible sugar content (7.6 g/L) as compare to other pre-treatments. Thermophilic fermentation of pre-treated *Anabaena* biomass by mixed bacterial culture was found suitable for H₂ production. Subhash and Venkata Mohan [59] applied an integrated biorefinery approach and used deoiled algal cake (after lipid extraction) as feed-stock for biohydrogen production using selectively enriched acidogenic consortia. Algae pre-treated extract (AP-E) documented maximum H₂ production rate (HPR), cumulative H₂ production (CHP) and specific H₂ yield (SHY) with higher substrate degradation (65%) in terms of COD removal efficiency than other conditions. The study signified the feasibility of microalgae as potential feedstock for simultaneous production of two energy forms viz., biodiesel and biohydrogen in the framework of biorefinery approach.

4. CONCLUSION

Fermentative hydrogen production from waste materials is considered as eco-friendly and promising route for hydrogen production. Fermentative bacteria can utilize a wide range of waste material as feedstock for H₂ production. The exploitation of waste materials for H₂ production not only leads to energy generation but also helps in bioremediation and in lowering the H₂ production cost, thus making the process economically viable. The major obstacle in H₂ production from waste material is the effective pre-treatment method of substrate, low H₂ rate and yield. To overcome the above obstacles, a lot of efforts and technical breakthroughs are required. A large variety of biomass and waters having different composition and origin have been tested as the potential substrate leading in many cases to very promising results. The food waste having high loading of carbohydrates have great potential for its use as substrate for biohydrogen production followed by most abundant and easily available lignocellulosic waste. Algal biomass has also shown good potential but production rate/yield is comparatively low. For efficient utilization of these materials, there is an urgent need to identify the effective, economically viable and nature- friendly methods for the hydrolysis/liberation of fermentable carbohydrates.

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5. REFERENCES

1. Das D, Veziroglu TN. Hydrogen production by biological processes: a survey of literature. Int J Hydrogen Energy. 2001; 26: 13-28.
2. Kapdan IK, Kargi F. Bio-hydrogen production from waste materials. Enzy Microbial Technol. 2006; 38: 569-582.

3. Rai PK, Singh SP. Biological production of clean energy: Hydrogen. In: Recent Advances in Microbiology S. P. Tiwari, Rajesh Sharma and Rajeeva Gaur, Editors. Nova Science Publishers Inc. New York, USA; 2013 p. 55-84.
4. Venkata Mohan S. Harnessing of biohydrogen from wastewater treatment using mixed fermentative consortia: process evaluation towards optimization. Int J Hydrogen Energy. 2009; 34: 7460-7474.
5. Wang Y, Wang H, Feng X, Wang X, Huang J. Biohydrogen production from cornstalk wastes by anaerobic fermentation with activated sludge. Int J Hydrogen Energy. 2010; 35: 3092-3099.
6. Fang HHP., Li Chenlin, Zhang Tong. Acidophilic biohydrogen production from rice slurry. Int J Hydrogen Energy. 2006;31:683-692.
7. Seifert K, Waligorska M, Laniecki M. Hydrogen generation in photobiological process from dairy waste water. Int J Hydrogen Energy. 2010; 35: 9624-9629.
8. Eroglu E, Gunduz U, Yucel M, Turker L, Eroglu I. Photobiological hydrogen production from olive mill wastewater as sole substrate sources. Int J Hydrogen Energy. 2004; 29: 163-171.
9. Rai PK, Singh SP, Asthana RK. Biohydrogen Production from cheese whey wastewater in a two-step anaerobic process. Appl Biochem Biotechnol. 2012; 167: 1540-1549.
10. Melis A, Zhang L, Forestier M, Ghirardi ML, Seifert M. Sustained photobiological hydrogen gas production upon reversible inactivation of oxygen evolution in the green alga *Chlamydomonas reinhardtii*. Plant Physiol. 2000; 122:127-133.
11. Hallenbeck PC, Ghosh D. Advances in fermentative biohydrogen production: the way forward. Trends Biotechnol. 2009; 27: 287-297.
12. Rai PK. Hydrogen production from dairy and agro wastes by integrating dark- and photo- fermentation. Ph.D thesis, Banaras Hindu University, Varanasi-22005, India. 2013.
13. Kim MS, Lee TJ, Yoon YS, Lee IG, Moon KW. Hydrogen production from food processing wastewater and sewage sludge by anaerobic dark fermentation combined with photofermentation. In: Miyake J, Matsunaga T, Pietro AS, editors. Biohydrogen II. Amsterdam: Elsevier 2001; p. 263-272
14. Cappelletti BM, Reginatto V, Amante ER, Antônio RV. Fermentative production of hydrogen from cassava processing wastewater by *Clostridium acetobutylicum*. Ren Energy. 2011; 36:3367-3372.
15. Rai PK, Singh SP, Asthana RK. Dairy waste based fermentative H₂ production. J Microbial World. 2011; 13: 207-213.
16. Rai PK, Singh SP, Asthana RK. Prospects of utilizing dairy waste for biohydrogen production. Int J Biotechnol Biosci. 2011; 1: 263-270.
17. Rai PK, Asthana RK, Singh SP. Optimization of photo- hydrogen production based on cheese whey spent medium. Int J Hydrogen Energy. 2014; 39: 7597-7603.
18. Xiao ND, Chen YG, Chen AH, Feng LY. Enhanced biohydrogen production from protein wastewater by altering protein structure and amino acids acidification type. Scientific Reports. 2014; 4, 3992; DOI: 10.1038/srep03992.
19. Jayalakshmi J, Joseph K, Sukumaran V. Bio hydrogen generation from kitchen waste in an inclined plug flow reactor. Int J Hydrogen Energy. 2009; 34:8854-8858.
20. Hwang JH, Choi JA, Abou-Shanab RAI, Min B, Song H, Kim Y, Lee ES, Jeon BH. Feasibility of hydrogen production from ripened fruits by a combined two-stage (dark/dark) fermentation system. Bioresour Technol. 2011; 102: 1051-1058.
21. Elbeshbishy E, Hafez H, Nakhla G. Ultrasonication for biohydrogen production from food waste. Int J Hydrogen Energy. 2011; 36: 2896-2903.
22. Zhang R, El Mashad HM, Hartman K, Wang F, Liu G, Choate C, Gamble P. Characterization of food waste as feedstock for anaerobic digestion. Bioresour Technol. 2007; 98: 929-939.
23. Lee DY, Ebie Y, Xu KQ, Li YY, Inamori Y. Continuous H₂ and CH₄ production from high-solid food waste in the two-stage thermophilic fermentation process with the recirculation of digester sludge. Bioresour Technol. 2010; 101: 542-547.
24. Kim DH, Kim SH, Shin HS. Hydrogen fermentation of food waste without inoculums addition. Enzyme Microb Technol. 2009; 45:181-87.
25. Kim SH, Shin HS. Effects of base- pretreatment on continuous enriched culture for hydrogen production from food waste. Int J Hydrogen Energy. 2008; 33:5266-5274.
26. Han SK, Shin HS. Biohydrogen production by anaerobic fermentation of food waste. Int J Hydrogen Energy. 2004; 29: 569-577.
27. Vijayaraghavan K, Ahmad D, Ibrahim MK. Biohydrogen generation from jackfruit peel using anaerobic contact filter. Int J Hydrogen Energy. 2006; 31:569-579.
28. Yasin HMN, Noraini AR, Hasfalina CM, Yusoff MZM, Hassan MA. Microbial characterization of hydrogen-producing bacteria in fermented food waste at different pH values. Int J Hydrogen Energy. 2011; 36: 9571-9580.
29. Danko AS, Pinheiro F, Abreu AA, Alves MM. Effect of methanogenic inhibitors, inocula type, and temperature on biohydrogen production from food components. Environ Eng Manage J. 2008; 7:531-536.
30. Elbeshbishy E, Hafez H, Dhar BR, Nakhla G. Single and combined effect of various pretreatment methods for biohydrogen production from food waste. Int J Hydrogen Energy. 2011; 36: 11379-11387.
31. Shin HS, Youn JH, Kim SH. Hydrogen production from food waste in anaerobic mesophilic and thermophilic acidogenesis. Int J Hydrogen Energy. 2004; 29:1355-1363.
32. Chu CF, Li YY, Xu KQ, Ebie Y, Inamori Y, Kong HN. A pH- and temperature-phased two-stage process for hydrogen and methane production from food waste. Int J Hydrogen Energy. 2008;33:4739-46.
33. Kim DH, Kim SH, Jung KW, Kim MS, Shin HS. Effect of initial pH independent of operational pH on hydrogen fermentation of food waste. Bioresour Technol. 2011; 102:8646-8652.
34. Venkata Mohan, Mohanakrishna G, Goud RK, Sarma PN. Acidogenic fermentation of vegetable based market waste to harness biohydrogen with simultaneous stabilization. Bioresour Technol. 2009; 100: 3061-68.
35. Kim DH, Kim SH, Kim KY, Shin HS. Experience of pilot-scale hydrogen producing anaerobic sequencing batch reactor (ASBR) treating food waste. Int J Hydrogen Energy. 2010; 35:1590-1594.
36. Reddy MV, Chandrasekhar K, Venkata Mohan S. Influence of carbohydrates and proteins concentration on fermentative hydrogen production using canteen based waste under acidophilic microenvironment. J Biotechnol. 2011; 155:387-395.
37. Kumar G, Bakonyi P, Periyasamy S, Kim SH, Nemestóthy N, Bélafi-Bakó K. Lignocellulose biohydrogen: practical challenges and recent progress. Ren Sustain Energy Reviews. 2015; 44, 728-737.
38. Van Wyk JPH, Mohulatsi M. Biodegradation of wastepaper by cellulase from *Trichoderma viride*. Bioresour Technol. 2003; 86:21-23.
39. Yang H, Guo L, Liu F. Enhanced bio-hydrogen production from corncob by a two-step process: dark- and photo-fermentation. Bioresour Technol. 2010; 101:2049-2052.
40. Siroth K, Chollakup R, Chotineeranat S, Piyachomkwan K, Christopher GO. Processing of cassava waste for improved biomass utilization. Bioresour Technol. 2000; 71:63-69.
41. Manish S, Banerjee R. Comparison of biohydrogen production processes. Int J Hydrog Energy. 2008; 33:279-286.
42. Phowan P, Danvirutai P. Hydrogen production from cassava pulp hydrolysate by mixed seed cultures: Effects of initial pH, substrate and biomass concentrations. Biomass Bioenergy. 2014; 64; 1-10.
43. Cheng J, Su H, Zhou J, Song W, Cen K. Hydrogen production by mixed bacteria through dark and photo fermentation. Int J Hydrogen Energy. 2011; 36: 450-457,
44. Rai PK, Singh SP, Asthana RK. Biohydrogen production from sugarcane bagasse by integrating dark- and photo-fermentation. Bioresour Technol. 2014; 152:140-146.
45. Kumar G, Sivagurunathan P, Chen Chin-Chao, Lin Chiu-Yue. Batch and continuous biogenic hydrogen fermentation of acid pretreated de-oiled *Jatropha* waste (DJW) hydrolysate. RSC Adv. 2016; 6:45482-91.
46. Kumar G, Sen B, Sivagurunathan P, Lin Chiu-Yue. High rate hydrogen fermentation of cello-lignin fraction in de-oiled *jatropha* waste using hybrid immobilized cell system. Fuel 2016; 182:131-140.

47. Kawaguchi H, Hashimoto K, Hirata K, Miyamoto K. H₂ production from algal biomass by a mixed culture of *Rhodobium marinum* A-501 and *Lactobacillus amylovorus*. *J Biosci Bioeng*. 2001; 91:277-282.
48. Kumar K, Roy S, Das D. Continuous mode of carbon dioxide sequestration by *C. sorokiniana* and subsequent use of its biomass for hydrogen production by *E. cloacae* IIT-BT 08. *Bioresour Technol*. 2013;145:116-122.
49. Roy S, Kumar K, Ghosh S, Das D. Thermophilic biohydrogen production using pre-treated algal biomass as substrate. *Biomass Bioenergy*. 2014; 61:157-166.
50. Nguyen MT, Choi SP, Lee J, Lee JH, Sim SJ. Hydrothermal acid pretreatment of *Chlamydomonas reinhardtii* biomass for ethanol production. *J Microbiol Biotechnol*. 2009; 19:161-166.
51. Nguyen TAD, Kim KR, Nguyen MT, Kim MS, Kim D, Sim SJ. Enhancement of fermentative hydrogen production from green algal biomass of *Thermotoga neapolitana* by various pretreatment methods. *Int J Hydrogen Energy*. 2010; 35:13035-13040.
52. Park JH, Yoon JJ, Park HD, Kim YJ, Lim DJ, Kim SH. Feasibility of biohydrogen production from *Gelidium amansii*. *Int J Hydrogen Energy*. 2011; 36:13997-14003.
53. Jung KW, Kim DH, Shin HS. Fermentative hydrogen production from *Laminaria japonica* and optimization of thermal pretreatment conditions. *Bioresour Technol*. 2011; 102:2745-2750.
54. Jung KW, Kim DH, Kim HW, Shin HS. Optimization of combined (acid + thermal) pretreatment for fermentative hydrogen production from *Laminaria japonica* using response surface methodology (RSM). *Int J Hydrogen Energy*. 2011; 36: 9626-9631.
55. Liu CH, Chang CY, Cheng CL, Lee DJ, Chang JS. Fermentative hydrogen production by *Clostridium butyricum* CGS5 using carbohydrate-rich microalgal biomass as feedstock. *Int J Hydrogen Energy*. 2012; 37: 15458-15464.
56. Park JH, Cheon HC, Yoon JJ, Park HD, Kim SH. Optimization of batch dilute-acid hydrolysis for biohydrogen production from red algal biomass. *Int J Hydrogen Energy*. 2013; 38: 6130-6136.
57. Liu H, Wang G. Fermentative hydrogen production from macroalgae *Laminaria japonica* using anaerobic mixed Bacteria. *Int J Hydrogen Energy*. 2014; 39: 9012-9017.
58. Nayak BK, Roy S, Das D. Biohydrogen production from algal biomass (*Anabaena* sp. PCC 7120) cultivated in airlift photobioreactor. *Int J Hydrogen Energy*. 2014; 39:7553-7560.
59. Subhash G and Venkata Mohan SV. Deoiled algal cake as feedstock for dark fermentative biohydrogen production: An integrated biorefinery approach. *Int J Hydrogen Energy*. 2014; 39: 9573-9579.
60. Ferchichi M, Crabbe E, Hintz W, Gill GH, Almadidy A. Influence of culture parameters on biological hydrogen production by *Clostridium saccharoperbutylacetoni* ATCC 27021. *World J Microbiol Biotechnol*. 2005; 21: 855-862.
61. Yang P, Zhang R, McGarvey JA, Benemann JR. Biohydrogen production from cheese processing wastewater by anaerobic fermentation using mixed microbial communities. *Int J Hydrogen Energy*. 2007; 32: 4761-4771.
62. Azbar N, Dokgöz FTC, Keskin T, Korkmaz KS, Syed HM. Continuous fermentative hydrogen production from cheese whey wastewater under thermophilic anaerobic conditions. *Int J Hydrogen Energy*. 2009; 34:7441-7447.
63. Oh SE, Logan BE. Hydrogen and electricity production from a food processing wastewater using fermentation and microbial fuel cell technologies. *Water Res*. 2005; 39: 4673-4682.
64. Davila-Vazquez G, Cota-Navarro CB, Rosales-Colunga LM, León-Rodríguez A, Razo-Flores E. Continuous biohydrogen production using cheese whey: improving the hydrogen production rate. *Int J Hydrogen Energy*. 2009; 34:4296-4304.
65. Amorim NCS, Alves I, Martins JS, Amorim ELC. Biohydrogen Production from Cassava Wastewater in an Anaerobic Fluidized Bed Reactor. *Brazilian J Chem Eng*. 2014; 31:603-612.
66. Han SK, Shin HS. Biohydrogen production by anaerobic fermentation of food waste. *Int J Hydrogen Energy*. 2004; 29:569-577.
67. Wang CH, Lin PJ, Chang JS. Fermentative conversion of sucrose and pineapple waste into hydrogen gas in phosphate-buffered culture seeded with municipal sewage sludge. *Process Biochem*. 2006; 41:1353-1358.
68. Doi T, Matsumoto H, Abe J, Morita S. Feasibility study on the application of rhizosphere microflora of rice for the biohydrogen production from wasted bread. *Int J Hydrogen Energy*. 2009; 34: 1735-1743.
69. Li SL, Lin JS, Wang YH, Lee ZK, Kuo SC, Tseng IC, Cheng SS. Strategy of controlling the volumetric loading rate to promote hydrogen-production performance in mesophilic-kitchen-waste fermentor and the microbial ecology analyses. *Bioresour Technol*. 2011; 102:8682-8687.
70. Patra S, Sangyoka S, Boonmee M, Reungsang A. Biohydrogen production from the fermentation of sugarcane bagasse hydrolysate by *Clostridium butyricum*. *Int J Hydrogen Energy*. 2008; 33:6058-6065.
71. Panagiotopoulos IA, Bakker RR, De Vrije T, Koukios EG, Classen PAM. Pretreatment of sweet sorghum bagasse for hydrogen production by *Caldicellulosiruptor saccharolyticus*. *Int J Hydrogen Energy*. 2010; 35:7738-7747.
72. Wang Y, Wang H, Feng X, Wang X, Huang J. Biohydrogen production from cornstalk wastes by anaerobic fermentation with activated sludge. *Int J Hydrogen Energy*. 2010; 35:3092-3099.
73. Nasirian N, Almassi M, Minaei S, Widmann R. Development of a method for biohydrogen production from wheat straw by dark fermentation. *Int J Hydrogen Energy*. 2011; 36:411-420.
74. Cui M and Shen J. Effects of acid and alkaline pretreatments on the biohydrogen production from grass by anaerobic dark fermentation. *Int J Hydrogen Energy*. 2012; 37:1120-1124.
75. Han H, Wei L, Liu B, Yang H, Shen J. Optimization of biohydrogen production from soybean straw using anaerobic mixed bacteria. *Int J Hydrogen Energy*. 2012; 37:13200-13208.
76. Lay CH, Sen B, Chen CC, Wu JH, Lee SC, Lin CY. Co-fermentation of water hyacinth and beverage wastewater in powder and pellet form for hydrogen production. *Bioresour Technol*. 2013; 135:610-615.
77. Zhao L, Cao GL, Wang AJ, Guo WQ, Ren HY, Ren NQ. Simultaneous saccharification and fermentation of fungal pre-treated cornstalk for hydrogen production using *Thermoanaerobacterium thermosaccharolyticum* W16. *Bioresour Technol*. 2013; 145:103-107.
78. Song ZX, Li XH, Li WW, Bai YX, Fan YT, Hou HW. Direct bioconversion of raw corn stalk to hydrogen by a new strain *Clostridium* sp. FS3. *Bioresour Technol*. 2014; 157:91-97.
79. Shi X, Jung KW, Kim DH, Ahn YT, Shin HS. Direct fermentation of *Laminaria japonica* for biohydrogen production by anaerobic mixed cultures. *Int J Hydrogen Energy*. 2011; 36:5857-5864.
80. Ho KL, Lee DJ, Su A, Chang JS. Biohydrogen from lignocellulosic feedstock via one-step process. *Int J Hydrogen Energy*. 2012; 37: 15569-74.
81. Yun YM, Jung KW, Kim DH, Oh YK, Shin HS. Microalgal biomass as a feedstock for biohydrogen production. *Int J Hydrogen Energy*. 2012; 37:15533-15539.
82. Ferreira AF, Marques AC, Batista AP, PASS Marques, Gouveia L, Silva CM. Biological hydrogen production by *Anabaena* sp – yield, energy and CO₂ analysis including fermentative biomass recovery. *Int J Hydrogen Energy*. 2012; 37:179-190.
83. Cheng J, Xia A, Liu Y, Lin R, Zhou J, Cen K. Combination of dark- and photo-fermentation to improve hydrogen production from *Arthrospira platensis* wet biomass with ammonium removal by zeolite. *Int J Hydrogen Energy*. 2012; 37: 13330-13337.
84. Xia A, Cheng J, Lin R, Lu H, Zhou J, Cen K. Comparison in dark hydrogen fermentation followed by photo hydrogen fermentation and methanogenesis between protein and carbohydrate compositions in *Nannochloropsis oceanica* biomass. *Bioresour Technol*. 2013; 138:204-213.
85. Shi X, Kim DH, Shin HS, Jung KW. Effect of temperature on continuous fermentative hydrogen production from *Laminaria japonica* by anaerobic mixed cultures. *Bioresour Technol*. 2013; 144:225-231.
86. Batista AP, Moura P, Marques PASS, Ortigueira J, Alves L, Gouveia L. *Scenedesmus obliquus* as feedstock for biohydrogen production by *Enterobacter aerogenes* and *Clostridium butyricum*. *Fuel*. 2014; 117:537-543.

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