

Optimization of *Justicia gendarussa* Burm.f. fermentation by *Aspergillus oryzae* based on total phenolic, total flavonoid, and antioxidant capacity responses

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ABSTRACT

This study aimed to determine the optimal fermentation conditions for *Justicia gendarussa* leaves to produce compounds with the highest antioxidant activity. Solid-state fermentation was conducted using *Aspergillus oryzae*, with treatment variables including incubation time, temperature, moisture content, and inoculum concentration optimized using Design-Expert v.13.0 software. Methanol was used to extract the fermented products, which were analyzed for total phenolic content (TPC), total flavonoid content (TFC), and antioxidant activity using the 2,2-diphenyl-1-picrylhydrazyl (DPPH) and ferric antioxidant power (FRAP) methods. The results revealed that fermentation generally increased TPC by approximately threefold and antioxidant activity while decreasing TFC by half. Correlation analysis indicated a negative relationship between inoculum concentrations. Antioxidant activity, as determined by the DPPH method, increased at higher fermentation temperatures, whereas the FRAP method showed higher antioxidant capacity at lower inoculum concentrations. The optimal fermentation conditions for achieving the highest TPC, TFC, and antioxidant activity were an incubation period of 14 d at 45°C, 80% moisture content, and an inoculum concentration of 20%.

1. INTRODUCTION

Indonesia's pharmaceutical raw material (PRM) supply is heavily reliant on imports, highlighting the need to explore local natural resources. *Justicia gendarussa* Burm.f. (gandarusa), a wild shrub with a rich history in traditional medicine, is a promising candidate for PRM development. Known for its pharmacological properties, including antioxidant, anti-inflammatory, anticancer, and hepatoprotective properties, gandarusa contains bioactive compounds, such as alkaloids, triterpenoids, tannins, phenols, and flavonoids, making it a valuable resource for pharmaceutical applications [1].

Optimizing the bioactivity of gandarusa in its natural state remains challenging, as its bioactive compounds are often present in bound or inactive forms. Previous study by Suminto *et al.* has demonstrated that solid-state fermentation (SSF) can enhance the bioactivity of

Syaefudin Suminto, Department of Biochemistry, Faculty of Mathematics and Natural Sciences, IPB University, Indonesia. E-mail: syaefudin01@apps.ipb.ac.id plant-derived compounds by increasing their bioavailability and bioactivity [2]. Numerous microorganisms, particularly fungi, have been widely employed as biocatalysts in SSF. Fungi are often favored in this process because of their lower water requirements than bacteria, which align with the minimal water usage inherent to this technique.

Several studies have reported that fungal species, such as *Aspergillus niger*, *Monascus* spp., *Saccharomyces cerevisiae*, and *Aspergillus oryzae*, are effective for SSF applications [3]. Among these, *A. oryzae* has distinct advantages. In addition to its widespread use in fermented foods and attesting to its safety, *A. oryzae* exhibits robust enzymatic activity that has been shown to enhance both the quantity and quality of secondary metabolites in a variety of food substrates and herbal matrices [3,4]. For instance, SSF employing *A. oryzae* has been reported to increase the total phenolic content (TPC) and antioxidant activity of rice bran, turmeric, and *Moringa oleifera* [5-7]. Similarly, fermentation with *A. oryzae* successfully improved the levels of quercetin and kaempferol in mulberry leaves, suggesting its potential to enhance plant-derived bioactive [8].

Although previous studies have demonstrated that *A. oryzae* can enhance the phenolic content and antioxidant properties of various herbs, comprehensive research on optimizing the fermentation

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conditions for gandarusa is lacking. To address this knowledge gap, this study aimed to optimize SSF conditions for gandarusa using *A. oryzae.* Because SSF involves complex microbial activities influenced by numerous environmental and technical factors, selecting the most critical parameters is important.

Previous studies have identified incubation time, temperature, moisture content, and inoculum concentration as critical parameters that contribute to metabolite production and enhancement of bioactivities in SSF [8-12]. In this study, these parameters were systematically investigated to determine the maximum TPC, flavonoid content, and antioxidant capacity. These findings are expected to provide a foundation for the development of gandarusa-based pharmaceutical raw materials, thereby contributing to the sustainability of Indonesia's pharmaceutical industry.

2. MATERIALS AND METHODS

2.1 Microorganism and Materials

The fungal isolate *A. oryzae* IPBCC was obtained from the IPB Culture Collection, IPB University, Indonesia, for use in the fermentation process. The primary material, *J. gendarussa* Burm.f. (gandarusa), was sourced from the Tropical Biopharmaca Research Center, IPB University, Indonesia.

2.2. Water Content Analysis of Gandarusa

The water content of gandarusa leaves was analyzed to determine the initial moisture content, which can influence the fermentation process. Porcelain dishes were preheated at 105°C in an oven for 30 min to remove residual moisture and then cooled in a desiccator for 30 min before weighing. A 2 g sample of powdered gandarusa leaves was placed in a porcelain dish and heated at 105°C for 3 h. After cooling in a desiccator, the sample was reweighed until a constant weight was achieved. Moisture content was calculated using following the equation (1):

$$Water \ content \ (\%) = \frac{Initial \ weight \ (g) - Final \ weight \ (g)}{Initial \ weight \ (g)} \times 100\%$$
(1)

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2.3. Preparation of A. oryzae Inoculum

Inoculum was prepared according to the method described by Suminto *et al.* [9]. *A. oryzae* cultures were grown on potato dextrose agar and incubated at 25°C for 7 d to promote spore formation. Spores were

collected using a cork borer (10 mm in diameter), forming medium discs that were transferred into culture tubes containing 20% glycerol solution. The culture tubes were stored at -20° C to maintain the viability of the inoculum.

2.4. Solid-State Fermentation of Gandarusa

Solid-state fermentation was conducted to enhance the content of bioactive compounds in gandarusa. Ten grams of powdered gandarusa leaves were placed in a 250 mL Erlenmeyer flask and mixed with 10 mL distilled water. The mixture was sterilized in an autoclave at 121°C for 15 min. The fermentation formulations were designed using Design-Expert software, incorporating four key variables: incubation time (3–14 d), temperature (30–45°C), moisture content (20–80%), and inoculum concentration (20–100%).

As summarized in Table 1, eleven formulations were generated with three central points at formulas 3, 5, and 6. The moisture content was adjusted by adding 2–8 mL of sterile water, and 20% moisture content was achieved with the addition of 2 mL of water. The inoculum concentration was controlled by adding *A. oryzae* discs with a 100% concentration equivalent to five discs per sample. The selection of this range (20–100%) was based on a preliminary investigation that suggested that lower fungal inoculum concentrations might result in inadequate colonization, whereas higher levels could potentially decrease fermentation efficiency due to competition for nutrients and limited oxygen availability [9,13]. Fermentation was performed under sterile conditions according to the experimental design, and each treatment was replicated thrice to ensure reliability.

2.5. Extraction of Fermented Gandarusa

Fermented gandarusa was extracted by maceration for 24 h to obtain the bioactive compounds. Methanol was used as the solvent with a leaf powder-to-solvent ratio of 1:10 (w/v). The extract was filtered to separate the filtrate from the residue and the filtrate was concentrated using a rotary evaporator at 40°C and 100 rpm. The concentrated extract was stored at 4°C to prevent degradation during the long-term storage. The extraction yield was calculated as a percentage, using the following equation (2):

$$Yield (\%) = \frac{Weight of dried extract}{Weight of simplisia} \times 100\%$$
(2)
powder × (1 – Moisture content)

Table 1: Fermentation formula and extract yields of gandarusa fermented with Aspergillus oryzae.

Formula		Experimental results			
	Incubation time (d)	Temperature (°C)	Moisture content (%)	Inoculum concentration (%)	Extract Yield (%)
1	3.0	40.0	80	20	7.72
2	14.0	30.0	80	20	5.53
3	8.5	37.5	50	60	8.43
4	3.0	30.0	80	100	7.76
5	8.5	37.5	50	60	7.83
6	8.5	37.5	50	60	9.23
7	14.0	45.0	20	20	7.69
8	14.0	45.0	80	100	9.70
9	3.0	45.0	20	100	10.21
10	3.0	30.0	20	20	5.74
11	14.0	30.0	20	100	5.08

2.6. Determination of Total Phenolic Content (TPC)

The TPC was measured using the Folin–Ciocalteu method. A 20 μ L aliquot of the fermented extract was mixed with 120 μ L of 10% Folin–Ciocalteu reagent in a 96-well microplate, incubated for 5 min of incubation at room temperature. Subsequently, 80 μ L of 10% Na₂CO₃ solution was added, and the mixture was incubated for 30 min. As a control, the TPC was also determined on the unfermented gandarusa. Absorbance was measured at 750 nm using a spectrophotometer. A standard curve was prepared with gallic acid concentrations of 0, 50, 75, 100, 150, 200, and 225 μ g/mL following the same procedure as the samples. TPC was expressed as milligram gallic acid equivalents per gram dry weight (mg GAE/g DW) and calculated using following equation (3):

$$C = \frac{c \times V}{m} \times DF \tag{3}$$

where:

C = Total phenolic content (mg GAE/g DW)

c = Concentration of total phenolic from standard curve (mg/L)

V = Volume of sample (L)

m = Weight of sample (g)

DF = Dilution factor

2.7. Determination of Total Flavonoid Content (TFC)

TFC was determined using the aluminum chloride (AlCl₃) method. In a 96-well microplate, 10 μ L of the extract was mixed with 60 μ L methanol, 10 μ L of 10% AlCl₃, 10 μ L of 1 M CH₃COOK, and 120 μ L distilled water. After 30 min of incubation at room temperature, absorbance was measured at 415 nm. As a control, the TFC was also determined on the unfermented gandarusa. A quercetin calibration curve (0, 100, 150, 200, 250, 300, 400, and 500 ppm) was prepared and TFC was expressed as milligram quercetin equivalents per gram dry weight (mg QE/g DW) using equation (4):

$$C = \frac{c \times V}{m} \times DF \tag{4}$$

where:

C = Total flavonoid concentration (mg QE/g DW)

c = Total phenolic concentration from the standard curve (mg/L)

V = Volume of sample (L)

m = Weight of sample (g)

DF = Dilution factor

2.8. Antioxidant Capacity by 2,2-Diphenyl-1-Picrylhydrazyl (DPPH) Method

DPPH scavenging activity was measured by adding 100 μ L of 125 μ M DPPH solution (in methanol) to 100 μ L of extract. After 30 min of incubation at room temperature in the dark, absorbance was measured at 517 nm. As a control, the antioxidant capacity by DPPH was also determined on the unfermented gandarusa. Trolox calibration standards (0–100 μ M) were used. Antioxidant activity was expressed as μ mol Trolox equivalents per gram of dry weight extract (μ mol TE/g DW) using equation (5):

$$C = \frac{c \times V}{m} \times DF \tag{5}$$

where:

 $C = Antioxidant activity (\mu mol TE/g DW)$

 $c = Concentration of sample from standard curve (\mu M trolox)$

V = Volume of sample volume (L) m = Weight of sample (g)

DF = Dilution factor

2.9. Antioxidant Capacity by Ferric Reducing Antioxidant Power (FRAP) Method

The FRAP assay was conducted following the method described by Batubara *et al.* [14]. FRAP reagent was prepared by mixing 10 mM 2,4,6-tris(2-pyridyl)-(S)-triazine (TPTZ) in 300 mM acetate buffer (pH 3.6), 20 mM ferric chloride, and 40 mM hydrochloric acid in a 10:1:1 ratio. Twenty microliters of extract were mixed with 280 μ L FRAP reagent and incubated at 37°C in the dark for 30 min. The absorbance was measured at 593 nm. As a control, the antioxidant activity by FRAP was also determined on the unfermented gandarusa. Trolox calibration standards (0–400 μ M) were used. Antioxidant activity was expressed as μ mol TE/g DW using following equation (6):

$$C = \frac{c \times V}{m} \times DF \tag{6}$$

where:

 $C = Antioxidant activity (\mu mol TE/g DW)$

c = Total phenolic concentration of the sample from the standard curve (mg/L)

V = Volume of sample (L)

m = Weight of sample (g)

DF = Dilution factor

2.10. Data Analysis

Experimental data, including TPC, TFC, and antioxidant capacities (DPPH and FRAP), were analyzed using one-way analysis of variance (ANOVA) with a 95% confidence level. The optimization of fermentation was performed using Design-Expert v.13.0 software. The optimal formulation was selected based on the highest desirability value.

3. RESULTS AND DISCUSSION

3.1. Water Content of Simplisia and Yield of Fermented Gandarusa

The water content of gandarusa simplisia was determined to enhance the efficiency of the extraction process, prevent the degradation of active compounds, and optimize the storage and isolation of bioactive compounds. The results showed that the average water content of gandarusa simplisia was $9.05 \pm 0.14\%$. Subsequently, 10 g of gandarusa simplisia was fermented using *A. oryzae* under specific conditions and extracted using methanol through a maceration method.

As summarized in Table 1, formula 9 produced the highest extract yield of 10.21%, whereas formula 11 yielded the lowest extract yield of 5.08%. These findings indicate that fermenting gandarusa for 3 days at 30°C, 20% moisture content, and 100% inoculum concentration resulted in the highest extract yield compared to the other fermentation conditions tested. The yield reflects the amount of compounds extracted using methanol through maceration. Previous studies have reported a positive correlation between extract yield and TPC, TFC, and antioxidant capacity [15]. However, other studies have also indicated that there is not always a positive correlation between the yield and bioactivity of compounds in plant extracts [16].

3.2. TPC and TFC of Fermented Gandarusa

The TPC of fermented gandarusa was determined using the Folin-Ciocalteu method, with gallic acid as the standard [17]. Gallic acid was selected because of its phenolic nature, simple structure, and ability to react efficiently with Folin-Ciocalteu reagent to form a measurable color complex at a specific wavelength [18]. TPC was expressed as mg GAE/g DW.

As summarized in Table 2, the TPC of unfermented gandarusa was 0.810 mg GAE/g DW, which was significantly lower than the TPC of fermented gandarusa, averaging 3.056 mg GAE/g DW. This result indicated that solid-state fermentation increased the TPC of gandarusa by approximately threefold. Similar increases in TPC following solid-state fermentation were reported by Cao *et al.* [19], who studied the effects of fermentation using *A. oryzae*, *A. niger*, and *Monascus purpureus* on *Apocynum venetum* L. Among these microorganisms, *A. niger* produced the highest increase in TPC.

Among the 11 fermentation formulas, formula 7 yielded the highest TPC at 5.226 mg GAE/g DW, whereas formula 11 exhibited the lowest TPC at 1.478 mg GAE/g DW. These results demonstrated that TPC in gandarusa can increase up to five-fold through solid-state fermentation with *A. oryzae* under optimal conditions: 14 d of incubation at 45°C, 20% moisture content, and 20% inoculum concentration.

Puspitasari *et al.* [7] and Akbari *et al.* [20] also reported TPC enhancement under different fermentation conditions. Puspitasari optimized the fermentation of *Moringa oleifera* seed flour with *A. oryzae* at 30°C for 3 days and inoculum volume of $1.25 \text{ mL} (7.5 \times 10^6 \text{ spores})$, while Akbari fermented corn bran using *Lactobacillus reuteri* for 3 days at 35°C, 60% moisture content, and an inoculum volume of 2 mL (2 × 10⁷ CFU). Similarly, Ozdemir *et al.* [21] found that solid-state fermentation of hazelnut oil cake with *A. oryzae* increased TPC, with optimal conditions of 4.6 d of incubation at 24.6°C, 69.8% moisture content, and inoculum ratio 10^7 spores/g dry matter of solid substrate (1.5×10^6 spores).

Table 2: TPC,	TFC, and	l antioxidant	capacity	of unfermer	nted and	fermented
gandarusa.						

Formula	TPC (mg GAE/g DW)	TFC (mg QE/g DW)	Antioxidant capacity (DPPH) (µmol TE/g DW)	Antioxidant capacity (FRAP) (μmol TE/g DW)
0	0.810	1.272	0.267	0.633
1	3.925	0.688	0.397	1.194
2	4.588	1.588	0.243	1.384
3	3.226	0.720	0.461	1.241
4	2.317	0.476	0.335	0.604
5	4.183	0.900	0.400	1.203
6	1.819	0.737	0.371	0.853
7	5.226	0.899	0.369	1.598
8	2.962	0.379	0.392	0.749
9	1.865	0.739	0.414	0.448
10	2.027	0.309	0.262	0.563
11	1.478	0.133	0.208	0.441

Formula 0: Unfermented gandarusa (negative control). DPPH: Antioxidant capacity measured by DPPH (2,2-diphenyl-1-picrylhydrazyl) method. FRAP: Antioxidant capacity measured by ferric reducing antioxidant power, TPC: Total phenolic content, TFC: Total flavonoid content, mg GAE/g DW: Milligram gallic acid equivalents per gram dry weight, mg QE/g DW: Milligram quercetin equivalents per gram dry weight In contrast, TFC decreased from 1.272 mg QE/g DW in unfermented gandarusa to an average of 0.688 mg QE/g DW after fermentation. Only formula 2 (14 d at 30°C, 80% moisture content, and 20% inoculum) showed a slight, non-significant increase to 1.588 mg QE/g DW. Similar TFC reductions have been reported in studies of green tea and *Saccharina japonica* fermentation by *A. oryzae* [22,23].

The increase in TPC and decline in TFC during gandarusa fermentation suggests metabolic transformations driven by *A. oryzae*. For example, Kim *et al.* [22] reported that gallotannin degradation in green tea releases gallic acid, increasing TPC but reducing TFC due to the breakdown of epigallocatechin-3-gallate (EGCG). The conversion of EGCG to phenolic compounds such as caffeoylquinic acid, kaempferol rutinoside, and 3-*p*-coumaroylquinic acid has been documented in previous studies. However, it is important to note that these specific phenolic derivatives were not quantified in the present study.

3.3. Antioxidant Capacity of Fermented Gandarusa using DPPH and FRAP Assays

Antioxidant capacity was evaluated using two approaches: The DPPH radical scavenging assay and FRAP assay. Trolox, a vitamin E analog, was employed as the standard for both methods because of its reliable reactivity with reagents, providing consistent and accurate data [24]. The results are expressed in micromoles of Trolox equivalents per gram of dry weight (µmol TE/g DW).

As shown in Table 2, the antioxidant capacity of the unfermented gandarusa was $0.267 \,\mu$ mol TE/g DW using the DPPH method and $0.633 \,\mu$ mol TE/g DW using the FRAP method. Solid-state fermentation with *A. oryzae* enhanced the antioxidant capacity, yielding average values of $0.350 \,\mu$ mol TE/g DW (DPPH) and $0.934 \,\mu$ mol TE/g DW (FRAP). Both methods demonstrated an increase in antioxidant capacity after fermentation.

The DPPH assay revealed that formula 3, which involved fermentation for 8.5 days at 37.5°C, 50% moisture content, and 60% inoculum concentration, exhibited the highest antioxidant capacity, reaching 0.435 µmol TE/g DW. In contrast, the FRAP assay identified formula 7 as having the highest antioxidant capacity among all the tested formulas, achieving 1.598 µmol TE/g DW. This indicated that fermentation for 14 days at 45°C, 20% moisture content, and 20% inoculum concentration resulted in optimal enhancement of antioxidant capacity when assessed using the FRAP method.

The differences in the results between the DPPH and FRAP methods can be attributed to their distinct reaction mechanisms and potential interference from fermentation metabolites [25]. The DPPH method measures the antioxidant activity based on the ability of compounds to donate hydrogen atoms, thereby neutralizing free radicals. In contrast, the FRAP method evaluates the reducing power of antioxidants by assessing their capacity to reduce Fe^{3+} to Fe^{2+} , which is more responsive to compounds with high redox potentials [26]. In addition, fermentation-derived metabolites, such as organic acids and peptides, may differentially affect these assays by either enhancing or inhibiting their respective reactions, leading to observed discrepancies [4,27].

After fermentation, antioxidant capacity measured by DPPH and FRAP methods showed a strong correlation with TPC, confirming that phenolic compounds are the primary contributors to antioxidant activity [28-30]. Although flavonoids also influence antioxidant activity, the observed decrease in TFC post-fermentation did not diminish antioxidant capacity. Instead, an increase in non-flavonoid phenolic compounds likely enhanced this effect.

Fable 3: P-values and percentage contribu	ation of fermentation factors o	n TPC, TFC, and antioxidant	capacity (DPI	PH and FRAP) respe	onses
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Fermentation factors	<i>P</i> -value				Contribution (%)			
	ТРС	TFC	DPPH	FRAP	TPC	TFC	DPPH	FRAP
A-Incubation time (d)	0.3450	0.1461	0.2104	0.1736	14.17	3.66	11.39	11.96
B-Temperature (°)	0.4529	0.4022	0.0585	0.2438	8.06	0.76	53.65	7.42
C-Moisture content (%)	0.4589	0.0533	0.3546	0.2513	7.81	11.74	4.91	7.09
D-Inoculum concentration (%)	0.1752	0.0219	0.8136	0.0513	60.11	30.00	0.25	49.97
AB	0.8599	0.0597	0.7464	0.9493	0.38	10.39	0.47	0.01
AC	0.7178	0.1355	0.7853	0.3262	1.63	4.03	0.33	4.62
BC	0.4321	N/A	N/A	N/A	9.98	N/A	N/A	N/A
AD	N/A	0.0184	0.6961	0.2459	N/A	35.98	0.70	7.32

N/A: Not available, TPC: Total phenolic content, TFC: Total flavonoid content, DPPH: 2,2-diphenyl-1-picrylhydrazyl, FRAP: Ferric antioxidant power

Phenolic compounds neutralize free radicals (e.g., OH^{\bullet} , NO_2^{\bullet} , and ONOOH) by donating hydrogen atoms [31]. Functional groups such as -CH₂COOH, -CH=CHCOOH, methoxy (-OCH₃), and hydroxyl (-OH) further boost antioxidant activity by lowering the bond dissociation enthalpy and facilitating electron transfer [32]. This mechanism underscores the critical role of phenolic compounds in the antioxidant systems.

3.4. Effect of Solid-State Fermentation Factors on Response Variables

To evaluate the effects of fermentation factors (incubation time [A], temperature [B], moisture content [C], and inoculum concentration [D]) on TPC, TFC, and antioxidant capacity (DPPH and FRAP methods), a correlation analysis was performed using Design-Expert software with a two-level fractional factorial design. ANOVA was used to select the appropriate model, with a significance level of P < 0.05 (indicating statistical significance) and a lack-of-fit F-value > 0.05 (indicating non-significant lack-of-fit), suggesting a good model fit [33]. In addition, the effects of fermentation factors were identified based on their percentage contributions to the response variables.

As shown in Table 3, model D (inoculum concentration) had the highest significant impact on the TFC, as indicated by a *P*-value of 0.0219 (P < 0.05). For TPC, inoculum concentration also had the greatest effect, contributing the highest percentage among the fermentation factors (60.11%). As illustrated in Figure 1, TPC was the highest at 20% inoculum concentration, whereas higher concentrations led to decreased phenolic content. Meanwhile, Figure 2 shows the effects of solid fermentation factors on TFC. The interaction between incubation time and inoculum concentration significantly influenced the TFC (P = 0.0184), with an insignificant lack-of-fit F-value (52.89), indicating model adequacy. The highest TFC was observed at 20% inoculum concentration, with lower inoculum concentrations and longer fermentation times tending to increase the TFC.

The effect of fermentation factors on the antioxidant capacity (DPPH) was assessed based on the percentage contribution and *P*-value. As shown in Table 3, temperature significantly affected DPPH with a percentage contribution of 53.65%. This was the highest among the fermentation factors, with a *P*-value of 0.0513. The highest DPPH antioxidant capacity was observed at 37.5°C (0.461 µmol TE/g DW), compared to 0.208 µmol TE/g DW at 30°C, indicating that higher fermentation temperatures enhanced antioxidant capacity [Figure 3].



Figure 1: Analysis of the effect of inoculum concentration on total phenolic content.

The effect of solid fermentation factors on the antioxidant capacity, measured using the FRAP method, is shown in Figure 4. Correlation analysis revealed that model D (inoculum concentration) contributed the highest percentage of the factors (49.97%). This result indicated that, at an inoculum concentration of 20%, the FRAP antioxidant capacity was the highest, reaching 1.598 µmoL TE/g DW. In contrast, at an inoculum concentration of 100%, FRAP antioxidant capacity decreased. This suggests that lower inoculum concentrations in solid fermentation led to a higher FRAP antioxidant capacity.

Overall, the correlation analysis identified inoculum concentration as the most influential fermentation factor affecting TPC, TFC, and FRAP antioxidant capacity. This supports previous studies showing that larger inoculum sizes increase the total phenolic and flavonoid content [34]. In addition to inoculum concentration, temperature and incubation time significantly influenced fermentation, particularly by promoting *A. oryzae* growth and metabolite production in the gandarusa. These findings align with those of Sulasiyah *et al.* [6] and Suminto *et al.* [9], who reported that the incubation time significantly affected metabolite production.



Figure 2: Analysis of the effect of inoculum concentration and incubation time on the response of total flavonoid content.



Figure 3: Analysis of the effect of temperature factor on the 2,2-diphenyl-1picrylhydrazyl antioxidant capacity response.

3.5. Optimization of Gandarusa Fermentation

The TPC, TFC, and antioxidant capacity (measured by DPPH and FRAP assays) were analyzed using the Design Expert software to determine the optimal fermentation conditions. Based on this analysis, a new formulation was identified by optimizing the ratios of the incubation time, incubation temperature, moisture content, and inoculum concentration. The selected solution achieved the highest



Figure 4: Analysis of the effect of inoculum concentration on ferric antioxidant power antioxidant capacity.

desirability value, where a value of 1 represents the optimum condition and values approaching 0 indicate undesired conditions.

In this study, the highest desirability value was 0.862, corresponding to optimal fermentation conditions of 14 days of incubation, fermentation temperature of 45°C, 80% moisture content, and 20% inoculum concentration. This formulation was predicted to yield the highest responses in terms of the TPC, TFC, and antioxidant capacity. Specifically, the predicted TPC was 4.726 mg GAE/g DW, TFC was 1.365 mg QE/g DW, and antioxidant capacities were 0.398 µmol TE/g DW (DPPH) and 1.644 µmol TE/g DW (FRAP).

Notably, our results indicate that the optimal fermentation temperature for A. oryzae, as suggested by the desirability value, exceeds the conventional optimal growth range of 32-36°C [35]. Although cultivation outside this range may adversely affect growth, recent research on cigar tobacco leaves has shown that higher temperatures (>45°C) can promote the establishment of microbial communities, including Aspergillus [36]. This finding suggests that fermentation temperature not only influences microbial proliferation but also shapes distinct community structures, ultimately affecting the production of physicochemical metabolites. Jiang et al. [37] reported that under high-temperature conditions (>42°C), genes involved in glucose, sucrose, and galactose metabolism in A. oryzae were downregulated. This downregulation may enhance the adaptability of fungi to elevated temperatures by reducing oxidative stress-related damage. In addition, the protease produced by A. oryzae exhibited thermal stability, with an optimum activity temperature of 57.2°C [38]. This is particularly noteworthy, as proteases can hydrolyze proteins and thereby release antioxidant compounds from protein-based substrates.

Furthermore, batch-to-batch variations may arise because of the inherent biological variability in fungal metabolism. Differences in the growth kinetics and enzymatic activity can affect the stability and production of bioactive compounds [39,40]. To address these

reproducibility concerns, implementing process standardization through consistent inoculum preparation, strict control of fermentation parameters, and continuous monitoring can help minimize batch variations and enhance overall process reliability [40,41].

4. CONCLUSION

Solid-state fermentation by *A. oryzae* has been shown to effectively increased the TPC and antioxidant capacity of gandarusa (*J. gendarussa* Burm. f). However, this fermentation process tends to reduce the TFC of the gandarusa. Correlation analysis and design solutions indicated that the optimal fermentation conditions for achieving the highest combined TPC, TFC, and antioxidant capacity were 14 days of fermentation at 45°C, 80% moisture content, and 20% inoculum concentration. Future research should involve an experimental verification to confirm the proposed optimal design.

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6. AUTHORS' CONTRIBUTIONS

All authors made substantial contributions to conception and design, acquisition of data, or analysis and interpretation of data; took part in drafting the article or revising it critically for important intellectual content; agreed to submit to the current journal; gave final approval of the version to be published; and agreed to be accountable for all aspects of the work. All the authors are eligible to be authors as per the International Committee of Medical Journal Editors (ICMJE) requirements and guidelines.

7. CONFLICTS OF INTEREST

The authors report no financial or any other conflicts of interest in this work.

8. ETHICAL APPROVALS

This study does not involve experiments on animals or human subjects.

9. DATA AVAILABILITY

All the data is available with the authors and shall be provided upon request.

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11. USE OF ARTIFICIAL INTELLIGENCE (AI)-ASSISTED TECHNOLOGY

The authors declares that they have not used artificial intelligence (AI)-tools for writing and editing of the manuscript, and no images were manipulated using AI.

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